Direct Bis-Arylation of Cyclobutanecarboxamide via Double C−H Activation: An Auxiliary-Aided Diastereoselective Pd-Catalyzed Access to Trisubstituted Cyclobutane Scaffolds Having Three Contiguous Stereocenters and an All-cis Stereochemistry

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S Supporting Information

ABSTRACT: An auxiliary-aided Pd-catalyzed highly diastereoselective double C−H activation and direct bis-arylation of methylene C(sp³)−H bonds of cyclobutanecarboxamides and the syntheses of several novel trisubstituted cyclobutanecarboxamide scaffolds having an all-cis stereochemistry are reported. Extensive screening of various auxiliaries and reaction conditions was performed to firmly establish the optimized reaction conditions required for effecting the mono- or double C−H arylation of cyclobutanecarboxamides. The auxiliary-attached cyclobutanecarboxamides 15a, 15g, and 15h, prepared from the auxiliaries such as, 8-aminoquinoline, 2-(methylthio)aniline, and N′,N′-dimethylethane-1,2-diamine were found to undergo an efficient direct bis-arylation. The Pd-catalyzed arylation reaction of N-(quinolin-8-yl)cyclobutanecarboxamide 15a with one equivalent or more of aryl iodides, afforded the corresponding bis-arylated cyclobutanecarboxamides 16a−y. Nevertheless, the Pd-catalyzed arylation of 15a with just 0.5 equiv of the aryl iodides 13a, 13b, 13e, and 13m, selectively gave the corresponding monoarylated cyclobutanecarboxamides 17a−17d. The Pd-catalyzed arylation of 15g or 15h with one equivalent or more of aryl iodides afforded the bis-arylated cyclobutanecarboxamides 19a−19c and 21a−21m, respectively. However, the Pd-catalyzed arylations of compounds 15g or 15h with just 0.5 equiv of aryl iodides were ineffective. The stereochemistry of compounds obtained in this work was unambiguously assigned from the X-ray structures of representative products.

■ INTRODUCTION

Cyclobutane is the second smallest, strained four-membered carbocyclic ring in the family of smallest rings (as the first member in the family is cyclopropane ring) and presents as a core unit of natural products (Figure 1), pharmaceutical agents, and synthetic compounds.^{1−3} Several cyclobutane natural products, especially, monoarylated and b[is](#page-1-0)-arylated cyclobutanecarboxamides exhibit a vari[ety](#page-21-0) of biological activities and medicinal properties. $1-17$ For example, incarvillateine has been traditionally used in treating rheumatism and relieving pain as an ancient Chinese c[ru](#page-21-0)[de](#page-22-0) medicine named as "Jiaohao".⁴ Biyouyanagin⁵ is exhibiting substantial activity against HIV and inhibiting cytokine production. Litt[or](#page-21-0)alisone⁶ is an active agent for increased N[G](#page-21-0)Finduced neurite outgrowth in PC12D cells. Furthermore, cyclobutanes isolated fr[om](#page-21-0) Piper nigram and Piper chaba are known to exhibit broad pharmacological activities.^{7,8} Various research groups have elegantly completed the total synthesis of several cyclobutane-based natural products by usi[ng](#page-21-0) different routes.8−¹⁷

It is believed that in Nature the cyclobutane natural products are co[ns](#page-21-0)t[ru](#page-22-0)cted via the direct coupling of the parent monomeric olefins with a very high degree of stereo- and regiocontrol. On the other hand, in laboratories, the direct $[2 + 2]$ photocycloaddition reaction involving two similar olefins or heterodimerizations of two distinct olefins has been one of the most desirable routes for assembling cyclobutanes (Figure 2).^{18,19} However, the experimental synthetic method has limitations such as the head-to-head or head-to-tail type ad[di](#page-2-0)ti[ons,](#page-22-0) homodimerization and E/Z isomerization of olefins, thereby leading to the uncontrolled production of a complex mixture of stereoisomers without the stereo- and regiocontrol.²⁰ Fortunately, the crystal engineering tactic and solid-state photochemistry experiments lead to the assembly of [a](#page-22-0) set of stereochemically defined cyclobutanes.^{18,19} Apart from these techniques, there exist only rare examples of direct heterodimerizations of two different olefins lea[ding](#page-22-0) to the formation of cyclobutanes with a high regio- and stereocontrol.^{21,22} Hence, the construction of substituted, especially the monoarylated or bis-arylated, cyclobutanecarboxamides with a hig[h de](#page-22-0)gree of regio- and stereocontrol is a demanding task in organic chemistry

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monoarylated (or) bis-arylated cyclobutane natural products

Figure 1. Representative cyclobutane natural products.

due to the fact that many of the bioactive cyclobutane natural products contain an aryl group as one of the substituents.

An efficient and commanding route for the construction of stereodefined, arylated cyclobutanecarboxamides would be the direct metal-catalyzed C−H arylation of the C(sp³)−H bond of cyclobutanes. However, when compared to the field dealing with the $\begin{bmatrix} 2 & + & 2 \end{bmatrix}$ photocycloaddition-based dimerization of olefins leading to substituted cyclobutanes with an unpredictable stereocontrol, the direct C−H functionalization of cyclobutanes is an underexplored topic of research. To the best of our knowledge there exist only two exceptional reports in this regard,16,17 nevertheless the C−H activation reactions, especially, the Pd-catalyzed C−H activation/functionalization of $C(sp^3)$ –[H](#page-22-0) bond is one of the potential topics of the present and future research since the direct C−H functionalization reactions are considered to be the attractive methods in terms of simplicity and atom economy.23−³⁴

Employing the pioneering Pd-catalyzed C−H arylation strategy develope[d](#page-22-0) [by](#page-22-0) Daugulis et al., recently, Baran's $group^{16,17}$ reported the synthesis of piperarborenines and pipercyclobutanamide A. Baran's group elegantly established and [optim](#page-22-0)ized the reaction condition for an effective monoarylation of 6a and 7a, which was a crucial step during the synthesis of cyclobutane natural products (Scheme 1).^{16,17} In this regard, Baran' group reported that the Pd-catalyzed C−H arylation of the substrates 6a and 7a gave the corr[es](#page-2-0)p[ondi](#page-22-0)ng monoarylated products 6c (52%) and 7c (54%).^{16,17} In contrast to these results, during an attempt to synthesize pipercyclobutanamide A via the mono C−H vinylation of 7a[, Ba](#page-22-0)ran's group obtained the bis(olefinated)cyclobutane 8b (50%) instead of the mono-olefinated product $\hat{\textbf{sc}}^{.\text{17}}$ Notably, this is the only example available on the "double C−H functionalization of cyclobutane" using a vinyl iodide as the [cou](#page-22-0)pling partner. Baran's group has stated that "the reason for the direct bis(olefination) is unclear, but it may be that the vinyl iodide 8a is smaller than the aryl iodides, thereby leading to a more facile second reaction".¹⁷

It is noteworthy to mention that Yu's group has obtained the bis-arylated product 11b along with the monoarylated p[rod](#page-22-0)uct 11a from the Pd-catalyzed C−H arylation of the smallest carbocyclic ring, cyclopropanecarboxamide 9. 27c Daugulis's group has showed the occurrence of the bis-arylation of cyclohexane ring and monoarylation of cyclop[ent](#page-22-0)ane ring in their groundbreaking report on an auxiliary-directed, palladiumcatalyzed arylation of $C(sp^3)$ -H bonds of various carboxylic acids.^{26b}

Recently, we reported $35a$ an auxiliary-enabled and Pdcatal[yzed](#page-22-0) C−H arylation reaction of methylene C(sp³)−H bond of the cyclopropanec[arbo](#page-22-0)xamide 12 using four equivalents of 1-iodo-4-methoxybenzene (13a) in toluene, which exclusively gave the monoarylated cyclopropanecarboxamide 14^{35b} as the single isomer via the C−H activation (Scheme 2). During our previous work,^{35b} by using the similar experimental conditions

popular ways of assembling cyclobutane derivatives

Scheme 1. Existing Examples Dealing with C−H Activation of Cyclobutane Baran group's recent works on the synthesis of cyclobutane-based natural products

adopted for the cyclopropanecarboxamide 12, curiously, we attempted the Pd-catalyzed, C−H activation reaction of N-

(quinolin-8-yl)cyclobutanecarboxamide (15a) with 1-iodo-4 methoxybenzene (13a) in toluene. Surprisingly, in contrast to

Scheme 2. Theme of This Work: Double C−H Activation and Direct Bis-Arylation of Cyclobutanecarboxamides

Table 1. Optimization of the Reaction Conditions for the C−H Arylation of 15a

	H 15a $(0.25$ mmol)	(1 mmol) 13b; $X = 1$ 13 $c: X = Br$ 13d; $X = CI$	PdL_2 (mol %) Oxidant (Y mmol) Solvent 80-110 °C		벖 C 16 _b	+	붜 17 _b
	entry catalyst $(mod \%)$	oxidant (Y mmol)	solvent (mL)	temp $(^{\circ}C)$	time(h)	16b: yield $(\%)^a$	17b: yield (%)
1	nil	AgOAc (1.0)	toluene	110	$\overline{24}$	$\overline{0}$	N. D.
$\overline{2}$	Pd(OAc) ₂ (5)	AgOAc (0.55)	1,2-DCE	80	15	68	N. D.
3	Pd(OAc) ₂ (5)	AgOAc (0.55)	1,4-Dioxane	100	15	63	N. D.
4	Pd(OAc) ₂ (5)	AgOAc (0.55)	CH ₃ CN	80	15	37	N. D.
5	Pd(OAc) ₂ (5)	AgOAc (0.55)	toluene	110	20	94	N. D.
6	$Pd(PPh_3)_{4}$ (5)	AgOAc (0.55)	toluene	110	15	0	N. D.
$\overline{7}$	$PdCl2$ (5)	AgOAc (0.55)	toluene	110	15	73	N. D.
8	$Pd(TFA)_{2}$ (5)	AgOAc (0.55)	toluene	110	15	70	N. D.
9	Pd(AcAc) ₂ (5)	AgOAc (0.55)	toluene	110	15	69	N. D.
10	$Pd(CH_3CN)_2Cl_2$ (5) AgOAc (0.55)		toluene	110	15	72	N. D.
11	Pd(OAc) ₂ (5)	$PhI(OAc)_2$ (0.55) toluene		110	24	5	N. D.
12	Pd(OAc) ₂ (5)	KOAc (0.55)	toluene	110	20	20	N. D.
13	Pd(OAc) ₂ (5)	$Cu(OAc)_{2}$ (0.55)	toluene	110	24	Ω	N. D.
14	Pd(OAc) ₂ (5)	Ag_2CO_3 (0.55)	toluene	110	24	68	N. D.
15 ^b	Pd(OAc) ₂ (5)	Ag_2CO_3 (0.55)	toluene	110	24	$\mathbf 0$	N. D.

a
All the reactions were done using phenyl iodide $(13b)$ and solvent (3 mL) under nitrogen atmosphere. The yields denoted here were calculated on The basis of the starting compound 15a. ^bIn this case, bromobenzene (13c) or chlorobenzene (13d) was used instead of iodobenzene (13b).

the Baran's substrate, $16,17$ we observed the formation of only the bis-arylated product 16a as the single diastereomer in 99% yield. Encouraged by this [outco](#page-22-0)me and in continuation of our interest on the construction of contiguous stereocenters of small carbocyclic rings using C−H activation route,³⁵ we envisaged

capitalizing on this result to accomplish the diastereoselective direct bis-arylation of cyclobutanes. In this regard, to the best of our knowledge there exist no reports dealing with the one-pot bis-arylation of the cyclobutane ring by directly using aryl iodides as the coupling partners.^{16,17,23−34}

a
All the reactions were done under nitrogen atmosphere. The yields denoted here were calculated based on the starting compound 15a.

Inspired by the latest developments on the Pd-catalyzed C−H activation reactions, we envisaged examining a direct protocol for the production of novel bis-arylated cyclobutanecarboxamide scaffolds under the 'auxiliary-aided and $Pd(OAc)_{2}$ -catalyzed C− H activation method'. 26b We herein, report an auxiliary-aided Pdcatalyzed highly diastereoselective "double C−H activation and direct bis-arylation [of m](#page-22-0)ethylene $C(sp^3)$ -H bonds of cyclobutanecarboxamides" and an efficient access to novel 1,2-cis, 1,3 cis, and 2,3-cis trisubstituted cyclobutane scaffolds having three contiguous stereocenters with a high degree of stereo- and regiocontrol.

■ RESULTS AND DISCUSSION

We commenced our study to find the optimized reaction conditions and solvents. Table 1 reveals the Pd- catalyzed arylation reaction of N -(quinolin-8-yl)cyclobutanecarboxamide (15a), obtained from cyclobutanecarbonyl chloride and an auxiliary, 8-aminoquinoline, with iodobenzene (13b). The reaction of N-(quinolin-8-yl)cyclobutanecarboxamide (15a) with iodobenzene (13b) in the absence of a palladium catalyst failed to afford any product (entry 1, Table 1). The C−H arylation of N-(quinolin-8-yl)cyclobutanecarboxamide (15a) in the presence of $Pd(OAc)_2$ catalyst and AgOAc i[n 1](#page-3-0),2-DCE gave the bis-arylated product 16b in 68% yield (entry 2, Table 1). Usage of other solvents such as 1,4-dioxane or MeCN did not improve the yield of the product 16b (entries 3 and 4, Table [1\)](#page-3-0).

The C−H arylation of cyclobutane 15a in the presence of $Pd(OAc)₂$ [an](#page-3-0)d AgOAc went smoothly in toluene at 110 °C and afforded the bis-arylated product 16b in 94% yield as a single diastereomer (entry 5, Table 1). Employing $Pd(PPh₃)₄$ as a catalyst failed to furnish the product 16b (entry 6, Table 1), and usage of other Pd catalysts ins[te](#page-3-0)ad of $Pd(OAc)$ ₂ also gave the product 16b in 69−73% yields (entries 7−10, Table 1). [Th](#page-3-0)e PdScheme 3. Double C−H Activation Route to Trisubstituted Cyclobutane Scaffolds Having Contiguous Stereocenters with an Allcis Stereochemistry a,b,c

 a All the reaction were done under nitrogen atmosphere. The yields denoted here were calculated based on the starting compound 15a. b The reaction was performed using 1.7 mol % (1 mg) of $Pd(OAc)_2$. The reaction was performed using 3 mol % (1.7 mg) of $Pd(OAc)_2$.

catalyzed arylation reaction of 15a with 13b in the presence of other oxidants/additives such as $PhI(OAc)₂$, KOAc, and $Cu(OAc)$ ₂ furnished the product 16b in significantly low yields or traces (entries 11−13, Table 1). The arylation of 15a with 13b in the presence of Ag_2CO_3 instead of AgOAc gave the bisarylated product 16b in 68[%](#page-3-0) yield (entry 14, Table 1). Employing bromobenzene (13c) or chlorobenzene (13d) instead of iodobenzene (13b) did not afford the product [16](#page-3-0)b (entry 15, Table 1).

In order to obtain products with very good yields and to have an efficient the C−H functionalization on the molecules, typically, more than one equivalent of aryl iodide (1−4 equiv) has been employed in the $C(sp^3)$ -H arylation strategy.²³⁻³⁵ However, though we have used the same experimental conditions and reagent amounts which were used for the [C](#page-22-0)−[H](#page-22-0) arylation of cyclopropanecarboxamides which gave only the monoarylated products, in this work, the C−H arylation of the cyclobutane 15a (1 equiv) with 1-iodo-4-methoxybenzene 13a

Scheme 4. Double C−H Activation Route to Trisubstituted Cyclobutane Scaffolds Having Contiguous Stereocenters with an Allcis Stereochemistry^a

 a All the reaction were done under nitrogen atmosphere. The yields denoted here were calculated based on the starting compound 15a.

(4 equiv) in the presence of $Pd(OAc)$ ₂ and AgOAc afforded only the bis-arylated product 16b as a single diastereomer (Scheme 2).

Further, the Baran's group also reported that the Pd-catalyzed [C](#page-3-0)−H arylation of 6a and 7a with excess of aryl iodides (6b and 7b, 2 equiv) gave only the corresponding monoarylated products 6c and $7c$ (Scheme 1).^{16,17} In order to have a clear insight regarding equivalents of phenyl iodides required for producing the monoarylated cy[clo](#page-2-0)[butan](#page-22-0)e product 17 or the bis-arylated cyclobutane product 16, we investigated the C−H arylation of cyclobutane 15a by varying the amount of the aryl iodide 13b (Table 2). Usage of one equivalent or more of iodobenzene (13b) in the Pd-catalyzed C−H arylation of 15a gave only the bis-aryl[ate](#page-4-0)d product 16b (entries 1−5, Table 2). The Pdcatalyzed C−H arylation of 15a with 0.75 equiv of iodobenzene (13b) gave both the bis-arylated cyclobutane 16b [\(11](#page-4-0)%) and the monoarylated cyclobutane 17b (33%) (entry 6, Table 2). Notably, the Pd-catalyzed C−H arylation reaction of 15a with just 0.5 equiv of iodobenzene (13b) in toluene (refluxed for [1](#page-4-0)5 h) selectively gave the monoarylated cyclobutane 17b in 20% yield (entry 7, Table 2). Increasing the reaction period from 15 to 48 h did not improve the reaction yield (entry 8, Table 2).

Next, we attempt[ed](#page-4-0) the Pd-catalyzed C−H arylation reaction of 15a with 0.5 equiv of iodobenzene by increasing the [ca](#page-4-0)talyst $(Pd(OAc)_2)$ loading to 15 mol %. However, this reaction also afforded the monoarylated cyclobutane 17b in 35% yield (entry 9, Table 2). Furthermore, we performed the Pd-catalyzed C−H arylation reaction of 15a by changing the reaction conditions and however[, o](#page-4-0)ur attempts to get the monoarylated cyclobutane 17b in good yields were not fruitful (entries 10−13. Table 2). Using the best reaction conditions (entries 7 and 9) described in the Table 2, the other monoarylated cyclobutane deriva[tiv](#page-4-0)es 17a, 17c and 17d were synthesized (Table 2).

Hav[in](#page-4-0)g the optimized reaction conditions in hand, the generality and scope of this auxiliary-a[id](#page-4-0)ed Pd-catalyzed double C−H activation and bis-arylation of the methylene $C(sp^3)$ −H

bond of cyclobutanecarboxamides were investigated (Scheme 3). A variety of substituted aryl iodides having valuable functional groups was used as the coupling partner in the Pd-cataly[ze](#page-5-0)d double C–H activation and bis-arylation of methylene C(sp³)– H bonds of cyclobutanecarboxamide 15a, thus offering an ample opening and an efficient access to novel 1,2-cis, 1,3-cis, and 2,3-cis trisubstituted cyclobutane scaffolds having three contiguous stereocenters with a high degree of stereo- and regiocontrol (Scheme 3).

The arylation of methylene C−H bonds of 15a with parasubstitut[ed](#page-5-0) aryl iodides having electron-withdrawing and -donating groups went smoothly and gave the corresponding cyclobutanes 16a, 16c−16i in 73−99% yields. The arylation of 15a with 1-iodonaphthalene, 1-iodo-3-methylbenzene, and 1 iodo-3-(trifluoromethyl)benzene gave the corresponding cyclobutanes 16j−16l in 59−99% yields. The arylation of 15a with meta-substituted aryl iodides having electron-withdrawing groups furnished the respective products 16m−16p in 80− 97% yields. The bis-arylation of the cyclobutanecarboxamide 15a with various disubstituted aryl iodides and 1-(4-iodophenyl) ethanone afforded the cyclobutane scaffolds 16q−16u in moderate to very good yields (Scheme 3). In some cases, we have performed the C−H arylation reaction of cyclobutane 15a using 3 mol % of the catalyst $Pd(OAc)₂$ $Pd(OAc)₂$ $Pd(OAc)₂$, which gave the corresponding bis-arylated products 16c and 16g in 97 and 85% yields, respectively (Scheme 3). Furthermore, we have attempted to use only 1.7 mol % (1 mg) of the catalyst $Pd(OAc)$ ₂ for the double C−H arylation of t[he](#page-5-0) cyclobutane derivative 15a. The C−H arylation of the cyclobutane 15a in the presence of just 1.7 mol % of $Pd(OAc)_2$ gave the corresponding bis-arylated products 16a (82%), 16c (81%), 16f (98%), 16k (68%), and 16q (72%) in good to moderate yields (Scheme 3).

Next, we aimed the auxiliary-aided Pd-catalyzed double C−H activation and bis-arylation of the methylene C(sp $^3)-\mathrm{H}$ bond of cyclobutanecarboxamides with various hete[ro](#page-5-0)aryl iodides (Scheme 4). The Pd-catalyzed double C−H functionalization

Scheme 5. Screening of Various Auxiliaries for the C−H Activation of Cyclobutane

and heterocyclic substitution on the cyclobutanecarboxamide 15a commendably gave the trisubstituted cyclobutanecarboxamide scaffolds 16v−16y in good to excellent yields (66−96%) (Scheme 4). It is noteworthy to mention that all these reactions selectively gave the 1,2-cis, 1,3-cis, and 2,3-cis trisubstituted cyclobut[an](#page-6-0)es having three contiguous stereocenters with a high degree of stereo- and regiocontrol, and the stereochemistries of the products 16a−16y (Schemes 3 and 4) were assigned on the basis of analysis of the X-ray structures of the representative compounds $16c$, 16f, 16g, and 1[6m](#page-5-0). 36

After the successful Pd-catalyzed bis-[ar](#page-6-0)ylation reactions of N- (quinolin-8-yl)cyclobutanecarboxam[id](#page-23-0)e (15a), we prepared a variety of cyclobutanecarboxamides 15b−15h by linking cyclobutane carbonyl chloride with various auxiliaries, respectively (Scheme 5). The Pd-catalyzed C−H arylation of cyclobutanecarboxamides 15b−15f did not afford the expected bis-arylated cyclobutanes 18a−18h. The reason for this may be that the corresponding auxiliaries linked with the cyclobutane ring have not aided the C−H functionalization of the cyclobutane ring.

Along this line, next we carried out the Pd-catalyzed arylation of N-(2-(methylthio)phenyl)cyclobutanecarboxamide (15g, prepared from an auxiliary, 2-(methylthio)aniline with different aryl iodides. Initially, we investigated the C−H arylation of cyclobutane 15g by varying the equivalents of 3-iodobenzaldehyde (13g, Table 3). Usage of one equivalent or more of 3 iodobenzaldehyde (13g) in the Pd-catalyzed C−H arylation of 15g gave only the bis-arylated product 19a (entries 1−3, Table 3). Similarly, subsequent examples of the bis-arylated cyclo-

Table 3. C−H Arylation N-(2- (Methylthio)phenyl)cyclobutanecarboxamide 15g

^aAll the reactions were done under nitrogen atmosphere. The yields denoted here were calculated based on the starting compound 15g. but the reaction was done using the reaction condition given for 'entry 1′.

Scheme 6. Trials Performed Aiming at the Monoarylation of 15a and 15g Using Baran's Reaction Condition^{a16,17}

^aThe yields denoted here were calculated based on the starting compound 15a or 15g.

butanes 19b and 19c having the 1,2-cis, 1,3-cis, and 2,3-cis stereochemistry were obtained in 45 and 29% yields, respectively (Table 3).³⁷ When compared to the starting material 15a, the C $-$ H arylation of 15g gave relatively low yields of the bis-arylated cyclob[ut](#page-7-0)a[ne](#page-23-0)s 19a−19c. However, unlike the substrate 15a, the Pd-catalyzed C−H arylation of 15g with just 0.5 equiv of 3 iodobenzaldehyde (13g) did not afford the monoarylated cyclobutane 20 (entry 4, Table 3).

Baran's group used the cyclobutanecarboxamides 6a and 7a which were similar to the cyclob[ut](#page-7-0)anecarboxamides 15g and 15a, respectively.16,17 When compared to this work, the Pd-catalyzed C−H arylation reaction conditions were different in the Baran's work. Baran'[s](#page-22-0) [gro](#page-22-0)up performed the C−H arylation of 6c and 7c with 2 equiv of 6b and 7b and revealed the formation of monoarylated products 6c and 7c, respectively. Contrastingly, under the experimental conditions of this work, the Pd-catalyzed C−H arylation of the substrates 15g or 15a with even just one equivalent of an aryl iodide gave only the bis-arylated cyclobutanes as the major compounds. Consequently, we were also interested to test the fate of our substrates 15g and 15a under Baran's reaction condition.^{16,17} Hence, we performed the C−H arylation reactions of the substrate 15a and 15g under the m[on](#page-22-0)oarylation reaction condition^{[16,1](#page-22-0)7} established by Baran's

group for the C−H arylation of 7a (Baran's substrate). However, our attempts resulted in the formation of only the bis-arylated cyclobutanes 16a, 16g, and 19b as the single isomers (Scheme 6) and we did not obtain the monoarylated cyclobutanes 17a, 17g, and 20b (Scheme 6). The reason for formation of only the respective bis-arylated compounds 16a, 16g, and 19b from 15a and 15g may be due to the substituent effect on the cyclobutane ring. Baran's substrates 6c and 7c have a substituent (carboxylic acid ester group) at the third position of cyclobutane ring, and hence the double arylation of 6c and 7c may not be a facile reaction due to the steric hindrance of aryl groups (if two aryl groups are introduced). However, the substrates investigated in this work, such as 15a, 15g, and 15h, do not have any substituent at the third position of the cycobutane ring when compared to Baran's substrate. Hence, the double arylations of cyclobutanecarboxamides 15a, 15g, and 15h occur in a facile manner.

Then, we scrutinized the scope of the Pd-catalyzed C−H activation and mono- or bis-arylation by using the N-(2- (dimethylamino)ethyl)cyclobutanecarboxamide (15h, prepared from an aliphatic auxiliary, N' , N' -dimethylethane-1,2-diamine and cyclobutanecarbonyl chloride). At the outset, we examined the Pd-catalyzed C−H arylation of N-(2-(dimethylamino) ethyl)-cyclobutanecarboxamide (15h) by varying the equivalents

Scheme 7. Investigation of the Pd-Catalyzed C−H Arylation of N-(2-(Dimethylamino)ethyl)-cyclobutanecarboxamide.^{a,b}

 a All the reactions were done under nitrogen atmosphere. The yields denoted here were calculated on the basis of the starting compound 15h. b The preparation of the compounds 21b−21m was carried out using the reaction condition given for 'entry 2′.

of 1-iodo-3-nitrobenzene (13f) (Scheme 7). The Pd-catalyzed C−H arylation of N-(2-(dimethylamino)ethyl)-cyclobutanecarboxamide (15h) by using either one equivalent or more of 1-iodo-3-nitrobenzene (13f) selectively afforded the bis-arylated cyclobutanecarboxamide 21a (entries 1−5, Scheme 7). These reactions did not afford any traces of the monoarylated cyclobutanecarboxamide 22a (entries 1−5, Scheme 7). However, unlike the substrate 15a, the Pd-catalyzed C−H arylation of 15h with just 0.5 equiv of iodo-3-nitrobenzene (13f) did not afford the monoarylated cyclobutane 22a (entry 6, Scheme 7). Further, several substituted aryl iodides and hetereoaryl iodides were used as the coupling partners in the Pd-catalyzed C−H activation and bis-arylation of methylene $C(sp^3)$ –H bonds of cyclobutanecarboxamide 15h, which led to the assembling of various 1,2-cis, 1,3-cis, and 2,3-cis trisubstituted cyclobutanecarboxamides 21a−21m (Scheme 7). Noticeably, all these reactions selectively gave the 1,2-cis, 1,3-cis, and 2,3-cis trisubstituted cyclobutanes having three contiguous stereocenters with a high degree of stereo- and regiocontrol and the stereochemistry of the products 21a−21m (Scheme 7) was assigned on the basis of the X-ray structure analysis of the representative compounds 21a and 21f.³⁶

Furthermore, we extended the scope this protocol by producing a wide range of novel tris[ub](#page-23-0)stituted cyclobutanecarboxamides 16z and 16aa−16ad having an all-cis stereochemistry by performing a second C−H activation/arylation on the corresponding monoarylated cyclobutanecarboxamides 17a and 17c (Scheme 8). The C−H functionalization reaction of the monoarylated cyclobutanecarboxamides 17a and 17c with a variety of hetereo[ary](#page-10-0)l iodides 13e, 13j, and 13k in the presence of

a
The overall yields were calculated for the reaction of the starting material 15a converting in to the corresponding bis-arylated products.

 $Pd(OAc)_2$ catalyst and AgOAc gave the corresponding trisubstituted cyclobutanecarboxamides 16z and 16aa−16ad in 30−82% yields, having two different aryl groups. The stereochemistry of the products 16z and 16aa−16ad (Scheme 8) was assigned on the basis of the X-ray structure analysis of the representative compounds 16c, 16f, 16g, and 16m and the similarity in the ${}^{1}H$ NMR pattern.

Next, we planned to construct a variety of trisubstituted cyclobutanecarboxamide frameworks analogous to the naturally occurring cyclobutanecarboxamide molecules shown in the Figure 1 (Scheme 9). In this line, initially, we performed the Pd-catalyzed C−H arylation reaction of N-(quinolin-8-yl)cyclobutane[ca](#page-1-0)rboxamide [\(](#page-11-0)15a) by using more than one equivalent of the aryl iodides 13l, 13m, 7b and 6b, which selectively afforded the corresponding bis-arylated cyclobutanecarboxamide frameworks 16ae, 16af, 16ag, and 16ah having an all cis-stereochemistry (Scheme 9).37 Along this line, the C−H functionalization reaction of the monoarylated cyclobutanecarboxamides 17a and 17d with the ar[yl](#page-11-0) i[od](#page-23-0)ides 6b and 13m in the presence of the catalyst $Pd(OAc)$, and the AgOAc gave the corresponding trisubstituted cyclobutanecarboxamide derivatives 16ai (40%) and 16aj (75%), which are structurally equivalent to some of the naturally occurring cyclobutanecarboxamide molecules with respect to the arene units (Scheme 9 and Figure 1).

Consequently, to reveal the synthetic value of this protocol, the Pd-catalyzed direct bis-arylatio[n](#page-11-0) of the C−[H](#page-1-0) bonds of the cyclobutanecarboxamide 15a was carried out in a gram scale, which afforded the product 16c in an excellent yield (95%) (Scheme 10). Additionally, we also tried to elaborate the Pdcatalyzed C−H activation of 15a by using different alkyl iodides

a.
The overall yields were calculated for the reaction of the starting material 15a converting in to the corresponding bis-arylated products.

Scheme 10. Gram Scale Double C−H Arylation of 15a

as the coupling partners. We performed the Pd-catalyzed C−H activation of 15a by using different alkyl iodides (13n−p) under several reaction conditions, however, all our attempts failed to

give the expected mono- or bis-alkylated compounds, such as 16ak, 16al, 16am, and 16an (Scheme 11).

In line with the pioneering studies carried out by Daugulis,^{26b} Chen^{28a} and a recent work by Charette,^{[24r](#page-12-0)} a plausible mechanism for the auxiliary-aided $Pd(OAc)_2$ -catalyzed, AgOAc-promo[ted](#page-22-0) doub[le C](#page-22-0)−H activation and direct b[is-a](#page-22-0)rylation of methylene $C(sp³)$ -H bonds of cyclobutanecarboxamides leading to the formation of monoarylated and bis-arylated cyclobutanecarboxamide is shown in the Scheme 12.

Additionally, to explore the synthetic utility we carried out the LiAlH4-mediated reduction of [the](#page-13-0) amide group of a representative compound 16g, which furnished $N-((1S*, 2R*, 4S^*)-2, 4-1)$ bis(4-bromophenyl)cyclobutyl)methyl)quinolin-8-amine (23). Further, the base-mediated amide hydrolysis of the representative compounds 16c, 16f, and 16g successfully gave the corresponding substituted bis-arylated cyclobutanecarboxylic acids 24−26 in very good yields, respectively (Scheme 13). The stereochemistry of the compounds 24−26 was unambiguously established based on the X-ray structure analysis [of](#page-13-0) a representative compound 25 , $36,38$ which revealed the occurrence of complete epimerization at the carbonyl group containing stereocenter of the cyclobut[anes](#page-23-0) 16c, 16f and 16g during the formation of the corresponding carboxylic acids 24−26 from the

Scheme 11. Trials on the Alkylation of Cyclobutanecarboxamide $15a^{a,b,c}$

 a All the reaction were done under nitrogen atmosphere. b In this case, 1.5 mmol of 13r was used. c The reaction was carried out in the open atmosphere.

base-mediated hydrolysis of the amides 16c, 16f and 16g. Treatment of the compound 16g with NaH followed by MeI gave the N-methylated cyclobutanamide derivative 27, having a cis-sterochemistry similar to the starting material 16g, and we did not observe epimerization in this reaction (Scheme 13). The stereochemistry of the product 27 was unambiguously assigned on the basis of the single-crystal X-ray structure analy[sis.](#page-13-0)³⁶

■ CONCLUSION

In summary, we have reported an auxiliary-aided Pd-catalyzed highly diastereoselective, double C−H activation and unprecedented direct bis-arylation of methylene C(sp³)-H bonds of cyclobutanecarboxamides. Extensive screening of several auxiliaries and reaction conditions was performed to firmly establish the regiocontrol and the exact reaction condition required for effecting the mono- or double C−H arylation of cyclobutanecarboxamides. The direct double C−H activation of cyclobutanecarboxamide led to the installation of two aryl groups on cyclobutanecarboxamide and a facile synthesis of several novel 1,2-cis, 1,3-cis and 2,3-cis trisubstituted cyclobutanecarboxamide frameworks having three contiguous stereocenters, 39 which are relatively difficult to prepare via the existing, popular direct $[2 +$ 2] photocycloaddition method with a high degree [of](#page-23-0) regio- and stereocontrol.

EXPERIMENTAL SECTION

General. Melting points are uncorrected. IR spectra were recorded as thin films or KBr pellets. 1 H and 13 C NMR spectra

Scheme 12. Plausible Mechanism for the Double C−H

were recorded on 400 and 100 MHz spectrometers respectively using CDCl₃ or DMSO- d_6 as solvent and TMS as an internal standard. HRMS measurements reported in this work were obtained from TOF and quadrupole mass analyzers. Column chromatography was carried out on silica gel (100−200 mesh). Reactions were carried out in anhydrous solvent under nitrogen atmosphere. Solutions were dried using anhydrous Na₂SO₄. Thin layer chromatography (TLC) was performed on silica gel plates and components were visualized by observation under iodine. Yields of products were not optimized. In all the reactions, the column chromatographic purification of the reaction mixture afforded only the bis-arylated cyclobutanecarboxamides as the major diastereomer in pure form. However, in special cases, monoarylated cyclobutanecarboxamides were obtained when the experimental condition was changed.

General Procedure for the Synthesis of Cyclobutanecarboxamides 15a−15c and 15e−15h. A dry flask containing the corresponding amine (auxiliary) (1 mmol), Et₃N (1.1 mmol) was stirred for 5−10 min under a nitrogen atmosphere. Then, to the reaction flask anhydrous DCM (4 mL) was added followed by dropwise addition of cyclobutanecarbonyl chloride (1 mmol). The resulting mixture was stirred for 10 min at rt. Then, the reaction mixture was refluxed for 12 h. After this period, the reaction mixture was diluted with dichloromethane and washed with water and twice with saturated aqueous $NaHCO₃$ solution. The combined organic layers were dried over anhydrous $Na₂SO₄$, concentrated in vacuum and purification of the resulting reaction mixture by column chromatography (silica gel, 100−200 mesh, EtOAc/hexanes =20:80) furnished the corresponding cyclobutanecarboxamides 15a−15c and 15e− 15h.

Procedure for Synthesis of Cyclobutanecarboxamide **15d.** 2-Picolinic acid (1.5 mmol) was dissolved in $S OCl₂$ (4 mmol) and stirred for 24 h at rt under a nitrogen atmosphere.

Scheme 12. Plausible Mechanism for the Double C-H
Arylation of Cyclobutanecarboxamide^{24r,26b,28a} Epimerization
Epimerization Epimerization

After this period, the reaction mixture was concentrated in vacuum and diluted with anhydrous DCM (3 mL) under nitrogen atmosphere. Then, the DCM solution containing 2 picolinoyl chloride was added to a separate flask containing cyclobutanamine (1 mmol) and $Et₃N$ (1.1 mmol) in anhydrous DCM (2 mL). Then, reaction mixture was stirred at rt for 10 min. Next, the reaction mixture was refluxed for 12 h. Then, the reaction mixture was further diluted with dichloromethane (5 mL) and washed with water followed by saturated aqueous $NaHCO₃$ solution. The combined organic layers were dried over anhydrous Na_2SO_4 , concentrated in vacuum and purification of the resulting reaction mixture by column chromatography (EtOAc/hexanes =20:80) furnished the cyclobutanecarboxamide 15d.

General Procedure for the Preparation of Bis-arylated Cyclobutanecarboxamides 16a−16aj/19a−19c/21a− 21m. A solution of cyclobutanecarboxamide 15 (0.25 mmol), Pd(OAc)₂ (2.8 mg, 0.0125 mmol (5 mol %)), aryl iodide (1 mmol) AgOAc (91.8 mg, 0.55 mmol) in anhydrous toluene (2−3 mL) was heated at an appropriate temperature (73−110 °C, see the corresponding Tables/Schemes for specific examples) for an appropriate time (8−24 h, see the corresponding Tables/

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Schemes for specific examples) under a nitrogen atmosphere. After the reaction period, the reaction mixture was diluted with EtOAc and concentrated in vacuum and purification of the resulting reaction mixture by column chromatography (silica gel, 100−200 mesh) furnished the corresponding bis-arylated cyclobutanecarboxamides 16a-16aj/19a-19c/21a-21m (see the corresponding Tables/Schemes for specific examples).

General Procedure for the Preparation of Monoarylated Cyclobutanecarboxamides 17a−17d. A solution of N-(quinolin-8-yl)cyclobutanecarboxamide (15a, 56 mg, 0.25 mmol), Pd $(OAc)_2$ (2.8 mg, 0.0125 mmol (5 mol %)), aryl iodide (0.125 mmol) AgOAc (91.8 mg, 0.55 mmol) in anhydrous toluene (2−3 mL) was heated at an appropriate temperature (73−110 °C, see the corresponding Tables/Schemes for specific examples) for an appropriate time (15−24 h, see the corresponding Tables/Schemes for specific examples) under nitrogen atmosphere. After the reaction period, the reaction mixture was diluted with EtOAc and concentrated in vacuum and purification of the resulting reaction mixture by column chromatography (silica gel, 100−200 mesh) furnished the corresponding monoarylated cyclobutanecarboxamides 17a− 17d (see the corresponding Tables/Schemes for specific examples).

Procedure for the Preparation 23 from 16g. To dry flask was added anhydrous THF (4 mL) and (1S*,2R*,4S*)-2,4 bis(4-bromophenyl)-N-(quinolin-8-yl)cyclobutanecarboxamide (16g, 0.25 mmol) and the reaction flask was cooled to 0 $\rm{^{\circ}C}$ under a nitrogen atmosphere. To this solution was added LiAlH₄ (1) mmol) in portions and was refluxed overnight. Then EtOAc (3 mL) and water (1−2 mL) were added sequentially. The resulting solution was extracted with EtOAc $(2 \times 15 \text{ mL})$ and dried over anhydrous $Na₂SO₄$, and the solvent was removed by rotary evaporation; product was purified by column chromatography on silica gel (EtOAc/hexane, 10:90), which afforded the product 24.

Procedure for the Hydrolysis of the Bis-arylated Cyclobutanecarboxamides 16c, 16f, and 16g. The corresponding bis-arylated cyclobutanecarboxamides 16c or 16f or 16g (0.25 mmol) and NaOH (6 mmol) in ethanol (3 mL) were heated at 80 °C for overnight. After this period, the reaction mixture was diluted with water and extracted with ether $(2 \times 10 \text{ mL})$, and then acidified with 1 N HCl to get a pH \approx 2. Extraction with ether $(2 \times 10 \text{ mL})$ and drying of the combined organic layers over $Na₂SO₄$ was followed by evaporation of the solvent in vacuum and gave the corresponding carboxylic acids 24−26.

Procedure for N-Methylation of 16q. To a dry flask was added a suspension of NaH in oil (0.75 mmol) and washed with hexane $(2 \times 2 \text{ mL})$. Then, to this reaction flask, anhydrous THF (3 mL) and $(1S^*2R^*4S^*)$ -2,4-bis(4-bromophenyl)-N-(quinolin-8-yl)cyclobutanecarboxamide (16g, 0.25 mmol) were added, and the reaction flask was cooled to 0 °C and allowed to stir for 15 min under a nitrogen atmosphere. To this solution was added MeI (1.5 mmol) in dropwise and the reaction mixture was stirred at rt for 12 h. Then, EtOAc (3 mL) and water (1−2 mL) were added sequentially. The resulting solution was extracted with EtOAc (2×10 mL) and dried over anhydrous Na₂SO₄, and the solvent was removed by under vacuum, and the crude reaction mixture was purified by column chromatography on silica gel (EtOAc/hexane 15:85), which afforded the product 27.

N-(Quinolin-8-yl)cyclobutanecarboxamide (15a). Analytical TLC on silica gel, 1:4 ethyl acetate/hexane $R_f = 0.70$; brown color solid; 221 mg, 98% yield; mp 62−64 °C; ¹ H NMR

 $(400 \text{ MHz}, \text{CDCl}_3)$: δ 9.70 (br s, 1H), 8.79 (d, 1H, J = 8.0 Hz), 8.70 (d, 1H, J = 4.0 Hz), 8.02 (d, 1H, J = 8.0 Hz), 7.47–7.30 (m, 3H), 3.37−3.29 (m, 1H), 2.50−2.41 (m, 2H), 2.30−2.23 (m, 2H), 2.22−1.87 (m, 2H); ¹³C NMR (100 MHz, CDCl₃): δ 173.6, 148.0, 138.2, 136.2, 134.4, 127.8, 127.2, 121.5, 121.3, 116.2, 41.3, 25.4, 18.1; FT-IR (KBr): 2988, 1698, 1598, 1351, 778 cm⁻¹; HRMS (ESI): m/z [M + H]⁺ calcd for C₁₄H₁₅N₂O: 227.1184; found 227.1189.

N-(Naphthalen-1-yl)cyclobutanecarboxamide (15b). Analytical TLC on silica gel, 1:4 ethyl acetate/hexane $R_f =$ 0.70; red color liquid; 164 mg, 73%yield; ¹H NMR (400 MHz, CDCl₃): δ 7.95−7.92 (m, 2H), 7.64 (br s, 1H), 7.56−7.51 (m, 3H), 7.29 (q, 1H, J = 8.0 Hz), 3.55–3.48 (m, 2H), 2.41–2.33 (m, 1H), 2.08 (q, 1H, J = 8.0 Hz), 1.89−1.80 (m, 3H); 13C NMR $(100 \text{ MHz}, \text{CDCl}_3): \delta$ 177.6, 134.8, 134.4, 131.0, 129.4, 128.7, 127.5, 127.2, 126.6, 125.4, 121.8, 41.5, 26.2, 25.0, 17.6 ; FT-IR (DCM): 2940, 1611, 1511, 1432, 810 cm[−]¹ ; HRMS (ESI): m/z $[M + H]^{+}$ calcd for $C_{15}H_{16}NO: 226.1231$; found 226.1226.

N-(Pyridin-2-yl)cyclobutanecarboxamide (15c). Analytical TLC on silica gel, 1:4 ethyl acetate/hexane $R_f = 0.70$; white color liquid; 132 mg, 75% yield; ¹H NMR (400 MHz, CDCl₃): δ 9.67 (br s, 1H), 8.15–8.06 (m, 2H), 7.51 (q, 1H, $J = 8.0$ Hz), $6.85-6.82$ (m, 1H), 3.07 (q, 1H, J = 8.0 Hz), 2.24–2.16 (m, 2H), 1.94 (t, 2H, J = 4.0 Hz), 1.74–1.67 (m, 2H); ¹³C NMR (100 MHz, CDCl₃): δ 174.1, 152.0, 147.0, 138.4, 119.3, 114.6, 40.4, 24.9, 18.0; FT-IR (DCM): 2850, 1601, 1530, 1421, 801 cm[−]¹ ; HRMS (ESI): m/z [M + H]⁺ calcd for C₁₀H₁₃N₂O: 177.1027; found 177.1029.

N-Cyclobutylpicolinamide (15d). Analytical TLC on silica gel, 1:4 ethyl acetate/hexane $R_f = 0.60$; red color liquid; 140 mg, 81% yield; ¹H NMR (400 MHz, CDCl₃): δ 8.66 (br s, 1H), 8.22 $(d, 2H, J = 8.0 \text{ Hz})$, 7.67 $(t, 1H, J = 8.0 \text{ Hz})$, 7.44–7.41 $(m, 1H)$, 4.62 (q, 1H, J = 8.0 Hz), 2.46–2.39 (m, 2H), 2.10–2.01 (m, 2H), 1.82−1.75 (m, 2H); ¹³C NMR (100 MHz, CDCl₃): δ 163.1, 149.8, 147.8, 137.5, 126.1, 122.3, 44.6, 31.2, 15.2; FT-IR (DCM): 2953, 1523, 1510, 1435, 798 cm⁻¹; HRMS (ESI): *m*/z [M + H]⁺ calcd for $C_{10}H_{13}N_2O$: 177.1027; found 177.1029.

N-(2-Methoxyphenyl)cyclobutanecarboxamide (15e). Analytical TLC on silica gel, 1:4 ethyl acetate/hexane $R_f = 0.70$; brown color liquid; 164 mg, 80% yield; ¹H NMR (400 MHz, CDCl₃): δ 8.38 (q, 1H, J = 8.0 Hz), 7.73 (br s, 1H), 6.95–6.84 $(m, 2H)$, 6.83–6.75 $(m, 1H)$, 3.74 $(s, 3H)$, 3.14 $(q, 1H, J = 8.0)$ Hz), 2.37−2.27 (m, 2H), 2.18−2.10 (m, 2H), 1.95−1.82 (m, 2H); ¹³C NMR (100 MHz, CDCl₃): δ 173.1, 147.8, 127.7, 123.4, 120.8, 119.6, 109.8, 55.5, 41.0, 25.2, 18.0; FT-IR (DCM): 2990, 1672, 1588, 1393, 810 cm[−]¹ ; HRMS (ESI): m/z [M + H]⁺ calcd for $C_{12}H_{16}NO_2$: 206.1181; found 206.1189.

N-(2-(Phenylthio)phenyl)cyclobutanecarboxamide (15f). Analytical TLC on silica gel, 1:4 ethyl acetate/hexane R_f = 0.70; yellow color liquid; 240 mg, 85% yield; ¹H NMR (400 $MHz, CDCl₃)$: δ 8.54 (d, 1H, J = 8.0 Hz), 8.20 (br s, 1H), 7.61 (q, 1H, J = 8.0 Hz), 7.46 (q, 1H, J = 8.0 Hz), 7.27−7.18 (m, 2H), 7.17−7.07 (m, 4H), 3.09−3.04 (m, 1H), 2.16−2.08 (m, 4H), 1.94−1.89 (m, 1H), 1.80−1.60 (m, 1H); 13C NMR (100 MHz, CDCl3): δ 173.2, 139.9, 136.9, 135.6, 131.1, 129.3, 126.9, 126.2, 124.1, 120.7, 119.5, 41.0, 25.1, 17.8; FT-IR (DCM): 2965, 1578, 1559, 1401, 815 cm[−]¹ ; HRMS (ESI): m/z [M + H]+ calcd for $C_{17}H_{18}NOS: 284.1109$; found 284.1104.

N-(2-Methylthio)phenyl)cyclobutanecarboxamide (15g). Analytical TLC on silica gel, 1:4 ethyl acetate/hexane R_f = 0.70; White color solid; 194 mg, 88% yield; mp 66–68 °C; ¹H NMR (400 MHz, CDCl₃): δ 8.36 (d, 1H, J = 8.0 Hz), 8.27 (br s, 1H), 7.47−7.45 (d, 1H, J = 8.0 Hz), 7.29 (q, 1H, J = 4.0 Hz), 7.06

 $(t, 1H, J = 8.0 \text{ Hz})$, 3.27 (q, 1H, J = 8.0 Hz), 2.45–2.36 (m, 2H), 2.35 (s, 3H), 2.35−2.26 (m, 2H), 2.06−1.91 (m, 2H); 13C NMR $(100 \text{ MHz}, \text{CDCl}_3)$: δ 173.3, 138.4, 132.9, 128.9, 125.1, 124.1, 120.4, 41.1, 25.4, 18.8, 18.2; FT-IR (KBr): 3322, 1621, 1532, 1413, 807 cm[−]¹ ; HRMS (ESI): m/z [M + H]⁺ calcd for $C_{12}H_{16}NOS: 222.0952$; found 222.0943.

N-(2-(Dimethylamino)ethyl)cyclobutanecarboxamide (15h). Analytical TLC on silica gel, 2.5:2.5 methanol/ethyl acetate R_f = 0.50; light-yellow color liquid; 136 mg, 80% yield; $^1\rm H$ NMR (400 MHz, CDCl₃): δ 3.43 (q, 2H, J = 4.0 Hz), 3.01 (q, 1H, J = 4.0 Hz), 2.61 (t, 2H, J = 8.0 Hz), 2.36 (s, 6H), 2.30–2.25 $(m, 2H)$, 2.14−2.12 $(m, 2H)$, 2.03−1.93 $(m, 2H)$; ¹³C NMR $(100 \text{ MHz}, \text{CDCl}_3)$: δ 175.4, 57.7, 44.4, 39.8, 36.0, 25.3, 18.2; FT-IR (DCM): 3290, 1603, 1557, 1485, 861 cm[−]¹ ; HRMS (ESI): m/z [M + H]⁺ calcd for C₉H₁₉N₂O: 171.1497; found 171.1499.

(1S*,2R*,4S*)-2,4-Bis(4-methoxyphenyl)-N-(quinolin-8-yl)cyclobutanecarboxamide (16a). Analytical TLC on silica gel, 1.5:3.5 ethyl acetate/hexane $R_f = 0.60$; yellow color solid; 108 mg, 99% yield; mp 146−149 °C; ¹ H NMR (400 MHz, CDCl₃): δ 9.43 (br s, 1H), 8.65 (q, 1H, J = 4.0 Hz), 8.30 (q, 1H, J $= 4.0$ Hz), 7.95 (q, 1H, $J_1 = 8.0$ Hz), 7.29–7.20 (m, 7H), 6.72– 6.69 (m, 4H), 4.02−3.97 (m, 1H), 3.95−3.90 (m, 2H), 3.61 (s, 6H), 3.44 (dd, 1H, J₁ = 12.0 Hz, J₂ = 8.0 Hz) 2.65–2.61 (m, 1H); ¹³C NMR (100 MHz, CDCl₃): δ 169.2, 157.9, 147.8, 138.2, 136.1, 134.3, 132.7, 128.2, 127.7, 127.2, 121.3, 120.9, 116.4, 113.5, 55.1, 54.8, 38.5, 30.5; FT-IR (KBr): 2936, 1689, 1612, 1519, 825 cm[−]¹ ; HRMS (ESI): m/z [M + H]⁺ calcd for $C_{28}H_{27}N_2O_3$: 439.2021; found 439.2019.

(1S*,2R*,4S*)-2,4-Diphenyl-N-(quinolin-8-yl)cyclo**butanecarboxamide (16b).**³⁹ Analytical TLC on silica gel, 1.5:3.5 ethyl acetate/hexane $R_f = 0.70$; brown color solid; 89 mg, 94% yield; mp 145−147 °C; ¹[H](#page-23-0) NMR (400 MHz, CDCl₃): δ 9.47 (br s, 1H), 8.67 (q, 1H, $J = 4.0$ Hz), 8.25 (q, 1H, $J = 4.0$ Hz), 7.97 (q, 1H, J = 4.0 Hz), 7.32−7.15 (m, 11H), 7.06−7.02 (m, 2H), 4.13–4.10 (m, 1H), 4.06–3.99 (m, 2H), 3.53 (dd, 1H, J_1 = 12.0 Hz, J_2 = 8.0 Hz), 2.71–2.68 (m, 1H); ¹³C NMR (100 MHz, CDCl3): δ 168.9, 147.8, 140.6, 138.2, 136.1, 134.2, 128.1, 127.7, 127.2, 127.0, 126.1, 121.3, 120.9, 116.4, 54.6, 39.1, 29.9; FT-IR (KBr): 3342, 1669, 1521, 1484, 993 cm⁻¹; HRMS (ESI): m/z [M $+ H$ ⁺ calcd for C₂₆H₂₃N₂O: 379.1810; found 379.1824.

(1S*,2R*,4S*)-N-(Quinolin-8-yl)-2,4-di-p-tolylcyclobutanecarboxamide (16c). Analytical TLC on silica gel, 1.5:3.5 ethyl acetate/hexane $R_f = 0.70$; light-yellow color solid; 100 mg, 99% yield; mp 156−160 °C; ¹ H NMR (400 MHz, CDCl₃): δ 9.45 (br s, 1H), 8.65 (q, 1H, J = 4.0 Hz), 8.32 (q, 1H, J = 4.0 Hz), 7.94 (q, 1H, J = 8.0 Hz), 7.29−7.20 (m, 7H), 6.98 (d, 4H, J = 8.0 Hz), 4.09−4.04 (m, 1H), 4.0−3.93 (m, 2H), 3.51 (dd, 1H, J₁ = 20.0 Hz, J₂ = 12.0 Hz), 2.67–2.64 (m, 1H), 2.15 (s, 6H); ¹³C NMR (100 MHz, CDCl₃): δ 169.1, 147.5, 138.2, 137.5, 136.1, 135.4, 134.3, 128.9, 127.6, 127.3, 126.9, 121.2, 120.8, 116.4, 54.6, 38.9, 30.2, 21.0; FT-IR (KBr): 3359, 1689, 1595, 1484, 791 cm[−]¹ ; HRMS (ESI): m/z [M + H]⁺ calcd for $C_{28}H_{27}N_2O$: 407.2123; found 407.2117.

(1S*,2R*,4S*)-2,4-Bis(4-ethylphenyl)-N-(quinolin-8 yl)cyclobutanecarboxamide (16d). Analytical TLC on silica gel, 1.5:3.5 ethyl acetate/hexane $R_f = 0.70$; white color solid; 106 mg, 98% yield; mp 136−138 °C; ¹H NMR (400 MHz, CDCl₃): δ 9.43 (br s, 1H), 8.68 (q, 1H, $J = 4.0$ Hz), 8.30 (q, 1H, $J = 4.0$ Hz), 7.98 (q, 1H, J = 4.0 Hz), 7.32–7.22 (m, 7H), 7.02 (d, 4H, J = 8.0 Hz), 4.11–4.08 (m, 1H), 4.04–3.97 (m, 2H), 3.53 (dd, 1H, J_1 = 20.0 Hz, $J_2 = 12.0$ Hz) 2.70–2.66 (m, 1H), 2.50 (q, 4H, $J = 8.0$ Hz), 1.08 (t, 6H, J = 8.0 Hz); ¹³C NMR (100 MHz, CDCl₃): δ

169.2, 147.7, 141.8, 138.2, 137.8, 136.1, 134.3, 127.6, 127.5, 127.2, 127.0, 121.3, 120.8, 116.4, 54.7, 39.0, 30.2, 28.5, 15.5; FT-IR (KBr): 3358, 1519, 1462, 1378, 825 cm⁻¹; HRMS (ESI): *m/z* $[M + H]^{+}$ calcd for $C_{30}H_{31}N_2O$: 435.2436; found 435.2436.

(1S*,2R*,4S*)-2,4-Bis(4-nitrophenyl)-N-(quinolin-8-yl) cyclobutanecarboxamide (16e). Analytical TLC on silica gel, 2:3 ethyl acetate/hexane $R_f = 0.50$; brown color solid; 86 mg, 74% yield; mp 182−184 °C; ¹H NMR (400 MHz, CDCl₃): δ 9.65 (br s, 1H), 8.73 (q, 1H, J = 4.0), 8.19 (q, 1H, J = 8.0), 8.08− 8.03 (m, 5H), 7.46−7.36 (m, 6H), 7.28 (q, 1H, J = 4.0 Hz), 4.35−4.30 (m, 1H), 4.20−4.13 (m, 2H), 3.60 (dd, 1H, $J_1 = 20.0$ Hz, $J_2 = 12.0$ Hz), 2.89–2.83 (m, 1H); ¹³C NMR (100 MHz, CDCl₃): δ 167.8, 148.2, 148.1, 146.4, 138.0, 136.5, 133.3, 127.8, 127.6, 127.1, 123.4, 121.9, 121.7, 116.4, 54.5, 38.6, 30.0; FT-IR (KBr): 3344, 1682, 1596, 1391, 840 cm[−]¹ ; HRMS (ESI): m/z [M $+ H$]⁺ calcd for C₂₆H₂₁N₄O₅: 469.1511; found 469.1500.

(1S*,2R*,4S*)-2,4-Bis(4-chlorophenyl)-N-(quinolin-8 yl)cyclobutanecarboxamide (16f). Analytical TLC on silica gel, 1.5:3.5 ethyl acetate/hexane $R_f = 0.60$; light-yellow color solid; 110 mg, 99% yield; mp 171−174 °C; ¹ H NMR (400 MHz, CDCl₃): δ 9.48 (br s, 1H), 8.69 (q, 1H, J = 4.0 Hz), 8.27 (q, 1H, J $= 4.0$ Hz), 8.03 (q, 1H, J = 4.0 Hz), 7.36–7.27 (m, 3H), 7.25– 7.20 (m, 4H), 7.20−7.11 (m, 4H), 4.10−4.05 (m, 1H), 3.99− 3.97 (m, 2H), 3.42 (dd, 1H, $J_1 = 20.0$ Hz, $J_2 = 12.0$ Hz), 2.70− 2.63 (m, 1H); ¹³C NMR (100 MHz, CDCl₃): δ 168.5, 147.9, 138.9, 138.1, 136.3, 133.8, 131.9, 128.4, 128.2, 127.7, 127.2, 121.5, 121.4, 116.5, 54.3, 38.4, 30.1; FT-IR (KBr): 2988, 1698, 1598, 1351, 778 cm[−]¹ ; FT-IR (KBr): 3350, 1683, 1596, 1486, 790 cm^{-1} ; HRMS (ESI): m/z $[M + H]^{+}$ calcd for $C_{26}H_{21}Cl_2N_2O$: 447.1030; found 447.1010.

(1S*,2R*,4S*)-2,4-Bis(4-bromophenyl)-N-(quinolin-8 yl)cyclobutanecarboxamide (16g). Analytical TLC on silica gel, 2:3 ethyl acetate/hexane $R_f = 0.60$; light-yellow color solid; 131 mg, 99% yield; mp 158−160 °C; ¹ H NMR (400 MHz, CDCl₃): δ 9.49 (br s, 1H), 8.69 (q, 1H, J = 4.0 Hz), 8.27 (q, 1H, J $= 4.0$ Hz), 8.03 (q, 1H, J = 4.0 Hz), 7.37–7.29 (m, 7H), 7.28– 7.15 (m, 4H) 4.10−4.05 (m, 1H), 3.96−3.90 (m, 2H), 3.42 (dd, 1H, J_1 = 20.0 Hz, J_2 = 12.0 Hz), 2.70–2.64 (m, 1H); ¹³C NMR $(100 \text{ MHz}, \text{CDCl}_3): \delta$ 168.5, 147.9, 139.5, 138.1, 136.3, 133.9, 131.1, 128.8, 127.7, 127.2, 121.5, 121.4, 120.1, 116.5, 54.2, 38.5, 30.1; FT-IR (KBr): 3349, 1687, 1524, 1323, 824 cm[−]1; HRMS (ESI): m/z [M + H]⁺ calcd for C₂₆H₂₁Br₂N₂O: 535.0020; found 534.9999.

(1S*,2R*,4S*)-2,4-Bis(4-fluorophenyl)-N-(quinolin-8 yl)cyclobutanecarboxamide (16h). Analytical TLC on silica gel, 1:4 ethyl acetate/hexane $R_f = 0.70$; pale-yellow color solid; 100 mg, 97% yield; mp 137−139 °C; ¹ H NMR (400 MHz, CDCl₃): δ 9.48 (br s, 1H), 8.69 (q, 1H, J = 4.0 Hz), 8.28 (q, 1H, J $= 4.0$ Hz), 8.02 (q, 1H, J = 8.0 Hz), 7.36–7.23 (m, 7H), 6.90– 6.89 (m, 4H), 4.10–4.05 (m, 1H), 4.00 (dd, 2H, J_1 = 20.0 Hz, J_2 = 12.0 Hz), 3.48 (dd, 1H, $J_1 = 20.0$ Hz, $J_2 = 12.0$ Hz), 2.71–2.67 (m, 1H); ¹³C NMR (100 MHz, CDCl₃): δ 168.7, 161.5 (d, J_{C−F} = 242 Hz), 147.9, 138.2, 136.3, 136.2 (d, J_{C−F} = 3 Hz), 134.0, 128.5 (d, J_{C-F} = 8.0 Hz), 127.7, 127.2, 121.4, 121.3, 116.4, 115.1 (d, $J_{\text{C-F}}$ = 21.1 Hz), 54.5, 38.3, 30.3; FT-IR (KBr): 3349, 1605, 1599, 1392, 737 cm[−]¹ ; HRMS (ESI): m/z [M + H]+ calcd for $C_{26}H_{21}F_{2}N_{2}O: 415.1621$; found 415.1611.

(1S*,2R*,4S*)-2,4-Bis(4-iodophenyl)-N-(quinolin-8-yl) cyclobutanecarboxamide (16i). Analytical TLC on silica gel, 2:3 ethyl acetate/hexane $R_f = 0.60$; brown color solid; 114 mg,73% yield; mp 169−171 $^{\circ}$ C; ¹H NMR (400 MHz, CDCl₃): δ 9.48 (br s, 1H), 8.70 (q, 1H, $J = 4.0$ Hz), 8.28 (q, 1H, $J = 4.0$ Hz), 8.06 (q, 1H, J = 4.0 Hz), 7.50–7.46 (m, 4H), 7.39–7.24 (m, 3H),

 7.04 (d, 4H, J = 8.0 Hz), 4.11–4.06 (m, 1H), 3.97–3.90 (m, 2H), 3.42 (dd, 1H, J₁ = 20.0 Hz, J₂ = 12.0 Hz), 2.70–2.63 (m, 1H); ¹³C NMR (100 MHz, CDCl₃): δ 168.4, 147.9, 140.2, 138.1, 137.1, 136.3, 133.9, 129.0, 127.7, 127.2, 121.5, 121.4, 116.5, 91.6, 54.2, 38.6, 29.9; FT-IR (KBr): 3346, 1681, 1520, 1485, 808 cm⁻¹; HRMS (ESI): m/z [M + H]⁺ calcd for C₂₆H₂₁I₂N₂O: 630.9743; found 630.9737.

(1S*,2R*,4S*)-2,4-Di(naphthalen-1-yl)-N-(quinolin-8 yl)cyclobutanecarboxamide (16j). Analytical TLC on silica gel, 2:3 ethyl acetate/hexane $R_f = 0.60$; brown color solid; 112 mg, 94% yield; mp 225−227 °C; ¹H NMR (400 MHz, CDCl₃): δ 9.03 (br s, 1H), 8.43 (dd, 1H, $J_1 = 8.0$ Hz, $J_2 = 4.0$ Hz), 8.23 (d, 2H, J = 8.0 Hz), 7.85 (dd, 1H, J_1 = 8.0 Hz, J_2 = 4.0 Hz), 7.77 (dd, $1H, J_1 = 8.0$ Hz, $J_2 = 4.0$ Hz), 7.70–7.64 (m, 4H), 7.58–7.50 (m, 4H), 7.48–7.38 (m, 2H), 7.35–7.31 (m, 2H), 7.16 (q, 1H, J = 4.0 Hz), 7.03–6.94 (m, 2H), 4.77–4.72 (m, 3H), 4.08 (dd, 1H, J_1 = 12.0 Hz, J_2 = 8.0 Hz), 2.91–2.84 (m, 1H); ¹³C NMR (100 MHz, CDCl3): δ 168.1, 147.2, 137.7, 135.8, 135.7, 133.7, 133.5, 131.8, 128.8, 127.1, 126.9, 126.7, 126.0, 125.4, 125.2, 125.0, 123.5, 120.8, 120.4, 115.7, 56.6, 37.9, 28.1; FT-IR (KBr): 3351, 1683, 1523, 1485, 780 cm[−]¹ ; HRMS (ESI): m/z [M + H]⁺ calcd for $C_{34}H_{27}N_2O: 479.2123$; found 479.2115.

(1S*,2R*,4S*)-N-(Quinolin-8-yl)-2,4-di-m-tolylcyclobutanecarboxamide (16k). Analytical TLC on silica gel, 1:4 ethyl acetate/hexane $R_f = 0.70$; light-yellow color solid; 100 mg, 99% yield; mp 98–100[°]°C; ¹H NMR (400 MHz, CDCl₃): δ 9.46 $(br s, 1H)$, 8.70 (q, 1H, J = 4.0 Hz), 8.28 (q, 1H, J = 8.0 Hz), 8.0 $(q, 1H, J_1 = 8.0 \text{ Hz})$, 7.35–7.23 (m, 3H), 7.23–7.04 (m, 6H), 6.85 (d, 2H, J = 8.0 Hz), 4.14–4.10 (m, 1H), 4.04–3.97 (m, 2H), 3.52 (dd, 1H, J_1 = 20.0 Hz, J_2 = 12.0 Hz), 2.71–2.67 (m, 6H), 2.13 (s, 6H); ¹³C NMR (100 MHz, CDCl₃): δ 169.1, 147.8, 140.6, 138.2, 137.5, 136.2, 134.3, 127.9, 127.7, 127.5, 127.2, 126.8, 124.0, 121.3, 120.9, 116.4, 54.6, 39.1, 30.0, 21.4; FT-IR (KBr): 3400, 1596, 1389, 1360, 1047 cm[−]¹ ; HRMS (ESI): m/z $[M + H]^{+}$ calcd for $C_{28}H_{27}N_{2}O: 407.2123$; found 407.2123.

 $(15*, 2R*, 45*)$ -N-(Quinolin-8-yl)-2,4-bis(3-(trifluoromethyl)phenyl)cyclobutanecarboxamide (16l). Analytical TLC on silica gel, 1:4 ethyl acetate/hexane $R_f = 0.70$; brown color solid; 75 mg, 59% yield; mp 114−116 °C; ¹H NMR $(400 \text{ MHz}, \text{CDCl}_3): \delta 9.46 \text{ (br s, 1H)}, 8.65 \text{ (q, 1H, } J = 4.0), 8.14$ $(q, 1H, J = 4.0 \text{ Hz})$, 7.99 $(q, 1H, J = 4.0 \text{ Hz})$, 7.50 $(s, 2H)$, 7.44 $(q,$ 2H, J = 8.0 Hz), 7.32−7.18 (m, 7H), 4.17−4.12 (m, 1H), 4.06− 3.99 (m, 2H), 3.51 (dd, 1H, $J_1 = 20.0$ Hz, $J_2 = 12.0$ Hz), 2.77– 2.69 (m, 1H); ¹³C NMR (100 MHz, CDCl₃): δ 168.2, 148.0, 141.3, 138.0, 136.2, 133.6, 130.3 (q, J_{C−F} = 32 Hz), 130.3, 128.5, 127.6, 127.1, 126.8 (q, $J_{C-F} = 271$ Hz), 123.7 (q, $J_{C-F} = 4$ Hz), 123.1 (q, J_{C-F} = 4 Hz), 121.4, 121.4, 116.3, 54.2, 38.7, 29.9; FT-IR (KBr): 3445, 1519, 1321, 1313, 989 cm⁻¹; HRMS (ESI): *m/z* $[M + H]^{+}$ calcd for $C_{28}H_{21}F_{6}N_{2}O$: 515.1558; found 515.1536.

(1S*,2R*,4S*)-2,4-Bis(3-nitrophenyl)-N-(quinolin-8-yl) cyclobutanecarboxamide (16m). Analytical TLC on silica gel, 2.5:2.5 ethyl acetate/hexane R_f = 0.50; brown color solid; 100 mg, 86% yield; mp 183−185 °C; ¹H NMR (400 MHz, CDCl₃): δ 9.57 (br s, 1H), 8.67 (q, 1H, $J = 4.0$ Hz), 8.16 (s, 2H), 8.16 (dd, 1H, $J_1 = 8.0$ Hz, $J_2 = 4.0$ Hz), 8.01 (dd, 1H, $J_1 = 8.0$ Hz, $J_2 = 4.0$ Hz), 7.88 (q, 2H, J = 4.0 Hz), 7.63 (d, 2H, J = 8.0 Hz), 7.36–7.28 $(m, 4H)$ 7.23 $(q, 1H, J = 4.0 Hz)$, 4.29–4.24 $(m, 1H)$, 4.17–4.10 $(m, 2H)$, 3.60 $(q, 1H, J = 12.0 Hz)$, 2.84–2.81 $(m, 1H)$; ¹³C NMR (100 MHz, CDCl₃): δ 167.9, 148.2, 148.1, 142.4, 138.0, 136.3, 133.4, 133.1, 129.0, 127.7, 127.0, 122.0, 121.7, 121.6, 121.4, 116.3, 54.2, 38.3, 29.7; FT-IR (KBr): 3344, 1682, 1579, 1485, 792 cm⁻¹; HRMS (ESI): m/z [M + H]⁺ calcd for $C_{26}H_{21}N_4O_5$: 469.1511; found 469.1512.

(1S*,2R*,4S*)-2,4-Bis(3-fluorophenyl)-N-(quinolin-8 yl)cyclobutanecarboxamide (16n). Analytical TLC on silica gel, 1:4 ethyl acetate/hexane $R_f = 0.70$; pale-yellow color solid; 100 mg, 97% yield; mp 132−134 °C; ¹ H NMR (400 MHz, CDCl₃): δ 9.49 (br s, 1H), 8.68 (q, 1H, J = 4.0 Hz), 8.25 (q, 1H, J $= 4.0$ Hz), 7.99 (q, 1H, J = 8.0 Hz), 7.33–7.20 (m, 3H), 7.12– 6.98 (m, 6H), 6.73−6.69 (m, 2H), 4.12−4.07 (m, 1H), 4.01− 3.93 (m, 2H), 3.45 (dd, 1H, $J_1 = 20.0$ Hz, $J_2 = 12.0$ Hz), 2.70− 2.63 (m, 1H); ¹³C NMR (100 MHz, CDCl₃): δ 168.4, 164.0 (d, J_{C-F} = 244 Hz) 147.9, 143.1 (d, J_{C-F} = 8.0 Hz), 138.1, 136.2, 133.9, 129.5 (d, J_{C-F} = 9.0 Hz), 127.7, 127.2, 122.5 (d, J_{C-F} = 3 Hz), 121.4, 121.3, 116.4, 114.1 (d, J_{C−F} = 21.5 Hz), 113.1 (d, J_{C−F} = 20.9 Hz); FT-IR (KBr): 3435, 1655, 1528, 1481, 793 cm[−]¹ ; HRMS (ESI): m/z [M + H]⁺ calcd for C₂₆H₂₁F₂N₂O: 415.1621; found 415.1609.

(1S*,2R*,4S*)-2,4-Bis(3-chlorophenyl)-N-(quinolin-8 yl)cyclobutanecarboxamide (16o). Analytical TLC on silica gel, 1:4 ethyl acetate/hexane $R_f = 0.70$; white color solid; 89 mg, 80% yield; mp 130−132 °C;¹H NMR (400 MHz, CDCl₃): δ 9.52 $(br s, 1H), 8.79 (q, 1H, J = 4.0 Hz), 8.30 (q, 1H, J = 4.0 Hz), 8.10$ $(q, 1H, J = 4.0 \text{ Hz}), 7.44-7.11 \text{ (m, 5H)}, 7.07-7.04 \text{ (m, 6H)},$ 4.19−4.14 (m, 1H), 4.06−4.0 (m, 2H), 3.50 (dd, 1H, $J_1 = 20.0$ Hz, $J_2 = 12.0$ Hz), 2.77–2.70 (m, 1H); ¹³C NMR (100 MHz, CDCl3): δ 168.3, 148.0, 142.5, 138.2, 136.2, 134.0, 133.8, 129.3, 127.7, 127.3, 127.2, 126.4, 125.1, 121.4, 121.3, 116.5, 54.3, 38.6, 29.8; FT-IR (KBr): 3434, 1601, 1525, 1323, 890 cm⁻¹; HRMS (ESI): m/z [M + H]⁺ calcd for C₂₆H₂₁NCl₂O: 447.1030; found 447.1018.

(1S*,2R*,4S*)-2,4-Bis(3-bromophenyl)-N-(quinolin-8 yl)cyclobutanecarboxamide (16p). Analytical TLC on silica gel, 2:3 ethyl acetate/hexane $R_f = 0.60$; yellow color liquid; 106 mg, 80% yield; ¹H NMR (400 MHz, CDCl₃): δ 9.55 (br s, 1H), 8.78 (q, 1H, J = 4.0 Hz), 8.33 (q, 1H, J = 4.0 Hz), 8.06 (q, 1H, J = 4.0 Hz), 8.50 (s, 2H), 7.40–7.21 (m, 7H), 7.08 (t, 2H, $J = 8.0$ Hz), 4.17–4.12 (m, 2H), 4.05–4.0 (m, 2H), 3.50 (dd, 1H, J_1 = 20.0 Hz, $J_2 = 12.0$ Hz), 2.74–2.71 (m, 1H); ¹³C NMR (100 MHz, CDCl3): δ 168.3, 148.0, 142.8, 138.1, 136.2, 133.8, 130.1, 129.7, 129.3, 127.7, 127.2, 126.4, 125.6, 122.4, 121.4, 121.3, 116.5, 54.3, 38.5, 29.9; FT-IR (DCM): 3332, 1592, 1423, 1361, 889 cm⁻¹; HRMS (ESI): m/z [M + H]⁺ calcd for C₂₆H₂₁Br₂N₂O: 535.0020; found 535.0024.

(1S*,2R*,4S*)-2,4-Bis(3,4-dimethylphenyl)-N-(quinolin-8-yl)cyclobutanecarboxamide (16q). Analytical TLC on silica gel, 1:4 ethyl acetate/hexane $R_f = 0.70$; light-yellow color solid; 107 mg, 99% yield; mp 99−102 °C; ¹H NMR (400 MHz, CDCl₃): δ 9.42 (br s, 1H), 8.64 (q, 1H, J = 4.0 Hz), 8.32–8.30 $(m, 1H)$, 7.92 (d, 1H, J = 8.0 Hz), 7.28–7.17 $(m, 3H)$, 7.08 (t, $4H, J = 8.0 \text{ Hz}$, 6.92 (d, 2H, $J = 8.0 \text{ Hz}$) $4.07 - 4.03 \text{ (m, 1H)}$, 3.92 $(q, 2H, J = 8.0 \text{ Hz})$, 3.48 (dd, 1H, $J_1 = 20.0 \text{ Hz}, J_2 = 12.0 \text{ Hz}$), 2.66−2.63 (m, 1H), 2.05 (s, 12H); 13C NMR (100 MHz, CDCl3): δ 169.4, 147.7, 138.2, 138.1, 136.1, 136.0, 134.4, 134.1, 129.4, 128.4, 127.6, 127.2, 124.5, 121.2, 120.8, 116.4, 54.7, 39.0, 30.3, 19.8, 19.4; FT-IR (KBr): 3435, 1575, 1365, 1291, 1047 cm⁻¹; HRMS (ESI): m/z [M + H]⁺ calcd for C₃₀H₃₁N₂O: 435.2423; found 435.2419.

(1S*,2R*,4S*)-2,4-Bis(3,5-dimethylphenyl)-N-(quinolin-8-yl)cyclobutanecarboxamide (16r). Analytical TLC on silica gel, 1:4 ethyl acetate/hexane $R_f = 0.70$; pale-yellow color liquid; 100 mg, 93% yield; $^1\rm H$ NMR (400 MHz, CDCl₃): δ 9.40 $(br s, 1H)$, 8.69 (q, 1H, J = 4.0 Hz), 8.28 (q, 1H, J = 4.0 Hz), 7.98 $(q, 1H, J = 4.0 \text{ Hz})$, 7.32–7.19 (m, 3H), 6.90 (s, 4H), 6.62 (s, 2H), 4.09–4.04 (m, 1H), 3.96–3.90 (m, 2H), 3.45 (dd, 1H, J_1 = 20.0 Hz, $J_2 = 12.0$ Hz), 2.67–2.62 (m, 1H), 2.09 (s, 12H); ¹³C

NMR (100 MHz, CDCl₃): δ 169.3, 147.7, 140.5, 138.3, 137.3, 136.1, 134.4, 127.8, 127.7, 127.2, 124.8, 121.2, 120.7, 116.4, 54.6, 39.0, 30.0, 21.3; FT-IR (DCM): 3358, 1690, 1520, 1387, 791 cm⁻¹; HRMS (ESI): m/z [M + H]⁺ calcd for C₃₀H₃₁N₂O: 435.2436; found 435.2423.

(1S*,2R*,4S*)-2,4-Bis(3,4-dichlorophenyl)-N-(quinolin-8-yl)cyclobutanecarboxamide (16s). Analytical TLC on silica gel, 1.5:3.5 ethyl acetate/hexane $R_f = 0.60$; white color solid; 51 mg, 40% yield; mp 140−145 °C; ¹H NMR (400 MHz, CDCl₃): δ 9.64 (br s, 1H), 8.81 (q, 1H, J = 4.0 Hz), 8.19 (q, 1H, J $= 8.0$ Hz), 8.09 (q, 1H, J = 4.0 Hz), 7.44–7.30 (m, 4H), 7.29– 7.18 (m, 5H), 4.58−4.53 (m, 1H), 4.19−4.12 (m, 2H), 3.60 (dd, 1H, J_1 = 20.0 Hz, J_2 = 12.0 Hz), 2.66–2.60 (m, 1H); ¹³C NMR $(100 \text{ MHz}, \text{CDCl}_3)$: δ 168.3, 148.0, 138.2, 136.3, 136.1, 134.0, 133.9, 132.6, 129.7, 128.8, 127.7, 127.1, 126.8, 121.4, 121.3, 116.2, 53.6, 37.4, 27.5; FT-IR (KBr): 3341, 1631, 1587, 1332, 804 cm^{-1} ; HRMS (ESI): m/z $[M + H]^{+}$ calcd for $C_{26}H_{19}Cl_4N_2O$: 515.0251; found 515.0249.

(1S*,2R*,4S*)-2,4-Bis(4-acetylphenyl)-N-(quinolin-8 yl)cyclobutanecarboxamide (16t). Analytical TLC on silica gel, 2:3 ethyl acetate/hexane $R_f = 0.30$; yellow color solid; 99 mg, 86% yield; mp181−183 °C; ¹H NMR (400 MHz, CDCl₃): δ 9.60 $(br s, 1H), 8.75 (q, 1H, J = 4.0 Hz), 8.25 (q, 1H, J = 4.0 Hz), 8.06$ $(q, 1H, J = 8.0 \text{ Hz})$, 7.81–7.79 (m, 4H), 7.42–7.24 (m, 7H), 4.28 (q, 1H, J = 4.0 Hz), 4.16–4.09 (m, 2H), 3.59 (dd, 1H, J₁ = 20.0 Hz, J_2 = 12.0 Hz), 2.83 (t, 1H, J = 4.0 Hz), 2.47 (s, 6H); ¹³C NMR $(100 \text{ MHz}, \text{CDCl}_3)$: δ 197.9, 168.4, 148.0, 146.3, 138.1, 136.3, 135.1, 133.8, 128.3, 127.7, 127.1, 127.0, 121.5, 121.4, 116.4, 54.4, 38.9, 29.8, 26.5; FT-IR (KBr): 3345, 1605, 1524, 1391, 793 cm[−]¹ ; HRMS (ESI): m/z [M + H]⁺ calcd for C₃₀H₂₇N₂O₃: 463.2021; found 463.2012.

(1S*,2R*,4S*)-2,4-Bis(4-methoxy-2-nitrophenyl)-N- (quinolin-8-yl)cyclobutanecarboxamide (16u). Analytical TLC on silica gel, 2:3 ethyl acetate/hexane $R_f = 0.30$; yellow color solid; 75 mg, 57% yield; mp145−147 °C; ¹ H NMR (400 MHz, CDCl₃): δ 9.53 (br s, 1H), 8.74 (q, 1H, J = 4.0 Hz), 8.12 (q, 1H, J $= 8.0$ Hz), 8.01 (q, 1H, $J = 8.0$ Hz), 7.57 (d, 2H, $J = 8.0$ Hz), 7.37−7.34 (m, 1H), 7.30−7.29 (m, 3H), 7.28−7.29 (m, 1H), 7.09−7.07 (m, 2H), 4.65 (q, 1H, J = 4.0 Hz), 4.34−4.27 (m, 2H), 3.70 (s, 6H), 3.52 (dd, 1H, $J_1 = 12.0$ Hz, $J_2 = 8.0$ Hz), 2.61 (q, 1H, $J = 8.0 \text{ Hz}$); ¹³C NMR (100 MHz, CDCl₃): δ 168.9, 158.2, 148.9, 148.2, 138.2, 135.8, 133.9, 130.8, 127.5, 126.7, 121.4, 121.2, 119.5, 116.1, 109.3, 55.6, 55.2, 35.9, 28.1; FT-IR (KBr): 3430, 1555, 1521, 1493, 819 cm[−]¹ ; HRMS (ESI): m/z [M + H]⁺ calcd for $C_{28}H_{25}N_4O_7$: 529.1723; found 529.1728.

(1S*,2R*,4S*)-2,4-Di(1H-indol-5-yl)-N-(quinolin-8-yl) cyclobutanecarboxamide (16v). Analytical TLC on silica gel, 2:3 ethyl acetate/hexane R_f = 0.30; yellow color semisolid; 75 mg, 66% yield; ¹H NMR (400 MHz, CDCl₃): δ 9.53 (br s, 1H), 8.95 $(br s, 2H)$, 8.45 (d, 1H, J = 4.0 Hz), 8.13 (d, 1H, J = 8.0 Hz), 7.90−7.78 (m, 1H), 7.66−7.39 (m, 2H), 7.22 (q, 1H, J = 4.0 Hz), 7.20−6.75 (m, 8H), 6.32 (s, 2H), 4.16 (t, 3H, J = 8.0 Hz), 3.58 (dd, 1H, $J_1 = 12.0$ Hz, $J_2 = 8.0$ Hz), 2.79 (t, 1H, $J = 8.0$ Hz); ¹³C NMR (100 MHz, CDCl₃): δ 169.9, 147.8, 138.1, 135.9, 134.6, 134.2, 131.5, 127.8, 127.5, 126.9, 124.3, 121.2, 120.7, 118.7, 116.2, 110.8, 101.8, 55.2, 39.7, 30.9; FT-IR (KBr): 3440, 1616, 1528, 1412, 710 cm[−]¹ ; HRMS (ESI): m/z [M + H]⁺ calcd for $C_{30}H_{25}N_4O: 457.2028$; found 457.2035.

(1S*,2R*,4S*)-N-(Quinolin-8-yl)-2-(thiophen-2-yl)-4- (thiophen-3-yl)cyclobutanecarboxamide (16w). Analytical TLC on silica gel, 1:4 ethyl acetate/hexane $R_f = 0.70$; yellow color solid; 81 mg, 83% yield; ¹H NMR (400 MHz, CDCl₃): δ 9.61 (br s, 1H), 8.75 (q, 1H, J = 4.0 Hz), 8.56 (t, 1H, J = 4.0 Hz), 8.08 (q,

 $1H, J = 4.0$ Hz), 7.40–7.37 (m, 3H), 7.10–7.06 (m, 4H), 6.92 (q, 2H, J = 4.0 Hz), 4.22−4.16 (m, 2H), 4.06−4.01 (m, 1H), 3.57 (dd, 1H, $J_1 = 20.0$ Hz, $J_2 = 12.0$ Hz), 2.95−2.88 (m, 1H); ¹³C NMR (100 MHz, CDCl₃): δ 168.1, 147.9, 143.4, 138.2, 136.2, 134.3, 127.7, 127.3, 126.7, 125.2, 123.9, 121.4, 121.2, 116.6, 55.9, 35.7, 35.5; FT-IR (DCM): 3313, 1612, 1554, 1302, 801 cm⁻¹; HRMS (ESI): m/z [M + H]⁺ calcd for C₂₂H₁₉N₂OS₂: 391.0938; found 391.0936.

(1S*,2R*,4S*)-2,4-Bis(6-fluoropyridin-3-yl)-N-(quinolin-8-yl)cyclobutanecarboxamide (16x). Analytical TLC on silica gel, 2:3 ethyl acetate/hexane $R_f = 0.30$; yellow color solid; 83 mg, 80% yield; mp187−189 ^oC; ¹H NMR (400 MHz, CDCl₃): δ 9.70 (br s, 1H), 8.76 (q, 1H, J = 4.0 Hz), 8.23–8.17 $(m, 4H)$, 7.86–7.81 $(m, 2H)$, 7.48 $(t, 2H, J = 4.0 Hz)$, 7.46–7.28 $(m, 1H)$, 6.80 $(q, 2H, J = 8.0 Hz)$, 4.20–4.05 $(m, 3H)$, 3.56 (dd, 1H, $J_1 = 20.0$ Hz, $J_2 = 12.0$ Hz), 2.79 (q, 1H, $J = 4.0$ Hz); ¹³C NMR (100 MHz, CDCl₃): δ 168.3, 162.4 (d, J_{C−F} = 236 Hz), 147.6, 146.2 (d, J_{C-F} = 14.5 Hz), 140.1 (d, J_{C-F} = 8 Hz), 137.8, 137.2, 133.1, 132.8, 128.0, 127.5, 122.3, 121.5, 118.2, 108.9 (d, J_{C-F} = 37.1 Hz), 54.1, 36.1, 30.0; FT-IR (KBr): 3430, 1624, 1554, 1341, 798 cm[−]¹ ; HRMS (ESI): m/z [M + H]+ calcd for C24H19F2N4O: 417.1526; found 417.1526.

(1S*,2R*,4S*)-2,4-Bis(2-chloropyridin-4-yl)-N-(quinolin-8-yl)cyclobutanecarboxamide (16y). Analytical TLC on silica gel, 2:3 ethyl acetate/hexane $R_f = 0.30$; yellow color liquid; 107 mg, 96% yield; ¹H NMR (400 MHz, CDCl₃): δ 9.70 (br s, 1H), 8.75 (q, 1H, J = 4.0 Hz), 8.22−8.15 (m, 3H), 8.10 (q, 1H, J $= 8.0$ Hz), 7.43–7.39 (m, 2H), 7.34–7.26 (m, 3H), 7.13–7.11 (m, 2H), 4.28−4.23 (m, 1H), 4.03−3.96 (m, 2H), 3.45 (dd, 1H, J_1 = 20.0 Hz, J_2 = 12.0 Hz), 2.78–2.73 (m, 1H); ¹³C NMR (100 MHz, CDCl₃): δ 167.3, 152.7, 151.5, 149.8, 148.3, 138.1, 136.5, 133.6, 127.8, 127.1, 122.7, 122.1, 121.7, 120.9, 116.7, 53.6, 37.7, 28.9; FT-IR (DCM): 3350, 1624, 1510, 1491, 810 cm⁻¹; HRMS (ESI): m/z [M + H]⁺ calcd for C₂₄H₁₉Cl₂N₄O: 449.0935; found 449.0936.

(1S*,2S*,4R*)-2-(4-Methoxyphenyl)-N-(quinolin-8-yl)- 4-(thiophen-2-yl)cyclobutanecarboxamide (16z). Analytical TLC on silica gel, 1:4 ethyl acetate/hexane $R_f = 0.70$; yellow color liquid; 69 mg, 67% yield; ¹H NMR (400 MHz, CDCl₃): δ 9.55 (br s, 1H), 8.75 (q, 1H, $J = 4.0$ Hz), 8.46 (q, 1H, $J = 4.0$ Hz), 8.09 (q, 1H, J = 4.0 Hz), 7.41–7.27 (m, 5H), 7.07–7.03 (m, 2H), 6.88−6.81 (m, 1H), 6.80−6.78 (m, 2H), 4.22−4.17 (m, 1H), 4.09−4.0 (m, 2H), 3.98 (s, 3H), 3.55 (dd, 1H, $J_1 = 12.0$ Hz, $J_2 =$ 8.0 Hz), 2.83–2.71 (m, 1H); ¹³C NMR (100 MHz, CDCl₃): δ 168.6, 157.9, 147.8, 143.8, 138.2, 136.2, 134.3, 132.4, 128.1, 127.7, 127.3, 126.6, 125.0, 123.8, 121.3, 121.0, 116.5, 113.5, 55.3, 55.1, 38.7, 35.2, 33.2; FT-IR (KBr): 3439, 1602, 1534, 1423, 990 cm⁻¹; HRMS(ESI): m/z [M + H]⁺ calcd for C₂₅H₂₃N₂O₂S: 415.1480; found 415.1488.

(1S*,2R*,4S*)-2-(1H-Indol-5-yl)-4-(4-methoxyphenyl)- N-(quinolin-8-yl)cyclobutanecarboxamide (16aa). Analytical TLC on silica gel, 2.5:2.5 ethyl acetate/hexane $R_f = 0.30$; yellow color liquid; 91 mg, 82% yield; ¹H NMR (400 MHz, CDCl₃): δ 9.60 (br s, 1H), 8.60 (q, 1H, J = 4.0 Hz), 8.32 (q, 1H, J $= 4.0$ Hz), 8.21 (br s, 1H), 7.96–7.94 (m, 1H), 7.67 (d, 1H, J = 4.0 Hz), 7.32−7.05 (m, 8H), 6.98−6.75 (m, 2H), 6.42−6.41 (m, 1H), 4.20−4.15 (m, 2H), 4.06 (t, 1H, J = 4.0 Hz), 3.68 (s, 3H), 3.59 (dd, 1H, J₁ = 12.0 Hz, J₂ = 8.0 Hz), 2.80–2.72 (m, 1H); ¹³C NMR (100 MHz, CDCl₃): δ 169.7, 157.8, 147.7, 138.2, 136.0, 134.6, 134.2, 132.9, 131.6, 128.2, 127.8, 127.6, 127.1, 124.2, 121.3, 121.2, 120.9, 118.8, 116.4, 113.5, 110.8, 102.1, 55.1, 55.0, 39.3, 38.7, 30.7; FT-IR (KBr): 3430, 1623, 1515, 1421, 890 cm⁻¹;

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HRMS(ESI): m/z [M + H]⁺ calcd for C₂₉H₂₆N₃O₂: 448.2025; found 448.2034.

(1S*,2R*,4S*)-2-(2-Chloropyridin-4-yl)-4-(4-methoxyphenyl)-N-(quinolin-8-yl)cyclobutanecarboxamide (16ab). Analytical TLC on silica gel, 2.5:2.5 ethyl acetate/hexane R_f = 0.30; light-yellow color liquid; 33 mg, 30% yield; ¹H NMR $(400 \text{ MHz}, \text{CDCl}_3): \delta 9.47 \text{ (br s, 1H)}, 8.70 \text{ (m, 1H)}, 8.22-8.04$ $(q, 1H, J = 4.0 Hz)$, 7.40–7.09 (m, 8H), 6.65 (q, 2H, J = 8.0 Hz), $4.24-3.87$ (m, 3H), 3.59 (s, 3H), 3.40 (dd, 1H, $J_1 = 20.0$ Hz, $J_2 =$ 12.0 Hz), 2.68–2.65 (m, 1H); ¹³C NMR (100 MHz, CDCl₃): δ 168.2, 158.1, 154.2, 151.4, 149.0, 148.0, 147.9, 144.3, 138.1, 136.3, 133.7, 131.3, 128.2, 127.7, 127.1, 122.6, 121.5, 120.9, 116.5, 113.5, 55.1, 54.2, 39.0, 37.3, 31.9; FT-IR (KBr): 3423, 1566, 1513, 1448, 810 cm[−]¹ ; HRMS (ESI): m/z [M + H]⁺ calcd for $C_{26}H_{23}CIN_{3}O_{2}$: 444.1478; found 444.1483.

(1S*,2S*,4R*)-2-(2-Chloropyridin-4-yl)-N-(quinolin-8 yl)-4-(thiophen-2-yl)cyclobutanecarboxamide (16ac). Analytical TLC on silica gel, 2.5:2.5 ethyl acetate/hexane $R_f =$ 0.30; yellow color solid; 54 mg, 52% yield; mp 182−184 °C; ¹H NMR (400 MHz, CDCl₃): δ 9.59 (br s, 1H), 8.77 (q, 1H, J = 4.0 Hz), 8.40 (q, 1H, J = 4.0 Hz), 8.24 (q, 1H, J = 4.0 Hz), 8.10 (q, 1H, $J = 8.0$ Hz), 7.44–7.35 (m, 3H), 7.28 (d, 1H, $J = 8.0$ Hz), 7.14 (q, 1H, J = 4.0 Hz), 7.03–7.02 (m, 2H), 6.82 (q, 1H, J = 8.0 Hz), 4.31−4.24 (m, 1H), 4.18−4.13 (m, 1H), 3.94−3.80 (m, 1H), 3.50 (dd, 1H, J_1 = 12.0 Hz, J_2 = 8.0 Hz), 2.89–2.82 (m, 1H); ¹³C NMR (100 MHz, CDCl₃): δ 167.7, 153.9, 151.5, 149.2, 148.0, 142.3, 138.2, 136.3, 133.8, 127.8, 127.3, 126.7, 125.4, 124.3, 122.5, 121.6, 121.5, 120.8, 116.6, 54.9, 37.5, 35.5, 32.5; FT-IR (KBr): 3530, 1654, 1529, 1409, 990 cm[−]¹ ; HRMS(ESI): m/z [M + H]⁺ calcd for C₂₃H₁₉ClN₃OS: 420.0937; found 420.0943.

(1S*,2S*,4R*)-2-(1H-Indol-5-yl)-N-(quinolin-8-yl)-4- (thiophen-2-yl)cyclobutanecarboxamide (16ad). Analytical TLC on silica gel, 2.5:2.5 ethyl acetate/hexane $R_f = 0.30$; red color liquid; 54 mg, 51% yield; ¹H NMR (400 MHz, CDCl₃): δ 9.60 (br s, 1H), 8.60 (q, 1H, J = 4.0 Hz), 8.40 (q, 1H, J = 4.0 Hz), 8.16 (br s, 1H), 8.01 (q, 1H, J = 8.0 Hz), 7.99 (d, 1H, J = 4.0 Hz), $7.65-7.23$ (m, 3H), $7.13-7.04$ (m, 5H), 6.88 (q, 1H, J = 4.0 Hz), 6.45 (q, 1H, J = 8.0 Hz), 4.29–4.10 (m, 3H), 3.62 (dd, 1H, J_1 = 12.0 Hz, J₂ = 8.0 Hz), 2.92–2.88 (m, 1H); ¹³C NMR (100 MHz, CDCl3): δ 169.1, 147.8, 144.0, 138.2, 136.0, 134.6, 134.3, 131.3, 127.9, 127.6, 127.2, 126.6, 125.0, 124.2, 123.8, 121.3, 121.2, 121.0, 118.7, 116.5, 110.8, 102.3, 55.6, 39.4, 35.5, 33.4; FT-IR (KBr): 3400, 1613, 1531, 1454, 910 cm[−]¹ ; HRMS(ESI): m/z [M $+ H$ ⁺ calcd for C₂₆H₂₂N₃OS: 424.1483; found 424.1492.

(1S*,2R*,4S*)-2,4-Bis(2,4-dimethoxyphenyl)-N-(quinolin-8-yl)cyclobutanecarboxamide (16ae). Analytical TLC on silica gel, 4:1 ethyl acetate/hexanes $R_f = 0.60$ red color liquid; 100 mg, 80% yield; ¹H NMR (400 MHz, CDCl₃): δ 9.52 (br s, 1H), 8.80 (q, 1H, J = 4.0 Hz), 8.36 (q, 1H, J = 4.0 Hz), 8.07 (q, 1H, J = 4.0 Hz), 7.40 (q, 1H, J = 4.0 Hz), 7.31–7.24 (m, 4H), 6.48 (q, 2H, J = 4.0 Hz), 6.25 (d, 2H, J = 4.0 Hz), 4.27–4.24 (m, 1H), 4.09−4.02 (m, 2H), 3.75 (s, 6H), 3.71 (s, 6H), 3.40 (dd, 1H, J_1 = 20.0 Hz, J_2 = 12.0 Hz), 2.64–2.60 (m, 1H); ¹³C NMR $(100 \text{ MHz}, \text{CDCl}_3)$: δ 170.3, 159.1, 158.0, 147.6, 138.2, 136.2, 134.8, 128.1, 127.7, 127.4, 122.1, 121.2, 120.2, 115.9, 103.5, 97.5, 55.3, 55.2, 53.8, 35.4; FT-IR (DCM): 2881, 1643, 1511, 1423, 811 cm⁻¹; HRMS (ESI): m/z [M + H]⁺ calcd for C₃₀H₃₁N₂O₅: 499.2333; found 499.2227.

(1S*,2R*,4S*)-2,4-Bis(3,4-dimethoxyphenyl)-N-(quinolin-8-yl)cyclobutanecarboxamide (16af). Analytical TLC on silica gel, 2.5:2.5 ethyl acetate/hexanes $R_f = 0.30$; pale-yellow color solid; 40 mg, 32% yield; mp 175−177 °C; ¹ H NMR (400 $MHz, CDCl₃)$: δ 9.56 (br s, 1H), 8.72 (q, 1H, J = 4.0 Hz), 8.43 (q, 1H, J = 4.0 Hz), 8.10 (q, 1H, J = 4.0 Hz), 7.42–7.28 (m, 3H), 6.90−6.87 (m, 4H), 6.76 (t, 1H, J = 8.0 Hz), 4.17−4.10 (m, 1H), $4.04-3.98$ (m, 2H), 3.73 (s, 6H), 3.73 (s, 6H), 3.72 (dd, 1H, J_1 = 20.0 Hz, $J_2 = 12.0$ Hz), 2.75–2.74 (m, 1H); ¹³C NMR (100 MHz, CDCl3): δ 169.3, 148.5, 147.7, 147.2, 138.2, 136.3, 134.2, 133.3, 127.7, 127.3, 121.3, 121.1, 119.1, 116.4, 110.8, 110.1, 50.7, 55.6, 54.6, 38.8, 31.1; FT-IR (KBr): 2922, 1601, 1545, 1434, 821 cm[−]¹ ; HRMS (ESI): m/z [M + H]⁺ calcd for C₃₀H₃₁N₂O₅: 499.2233; found 499.2245.

(1S*,2R*,4S*)-2,4-Bis(benzo-[d][1,3]-dioxol-5-yl)-N- (quinolin-8-yl)cyclobutanecarboxamide (16ag). Analytical TLC on silica gel, 4:1 ethyl acetate/hexanes $R_f = 0.70$ yellow color liquid; 58 mg, 50% yield; ¹H NMR (400 MHz, CDCl₃): δ 9.51(br s, 1H), 8.76 (q, 1H, J = 4.0 Hz), 8.41 (q, 1H, J = 4.0 Hz), 8.09 (q, 1H, J = 4.0 Hz), 7.44–7.33 (m, 3H), 6.85–6.79 (m, 4H), 6.68 (q, 2H, J = 4.0 Hz), 5.82 (d, 4H, J = 4.0 Hz), 4.09–4.04 (m, 1H), 3.98–3.91 (m, 2H), 3.42 (dd, 1H, $J_1 = 20.0$ Hz, $J_2 = 12.0$ Hz), 2.70−2.63 (m, 1H); ¹³C NMR (100 MHz, CDCl₃): δ 168.9, 147.8, 147.4, 145.8, 138.2, 136.2, 134.4, 134.2, 127.7, 127.7, 127.3, 121.3, 121.0, 120.0, 116.4, 107.9, 107.7, 100.7, 54.7, 38.8, 30.6; FT-IR (DCM): 3041, 1723, 1487, 1421, 991 cm⁻¹; HRMS (ESI): m/z [M + H]⁺ calcd for C₂₈H₂₃N₂O₅: 467.1607; found 467.1608.

(1S*,2R*,4S*)-N-(Quinolin-8-yl)-2,4-bis(3,4,5-trimethoxyphenyl)cyclobutanecarboxamide (16ah). Analytical TLC on silica gel, 2.5:2.5 ethyl acetate/hexanes $R_f = 0.30$ lightyellow color liquid; 70 mg, 50% yield; ¹H NMR (400 MHz, CDCl₃): δ 9.58 (br s, 1H), 8.73 (q, 1H, J = 4.0 Hz), 8.46 (q, 1H, J $= 4.0$ Hz), 8.12 (q, 1H, J = 4.0 Hz), 7.44–7.34 (m, 3H), 6.53 (s, 4H), 4.16 (q, 1H, J = 4.0 Hz), 4.03−3.91 (m, 2H), 3.73 (s, 12H), 3.67 (s, 6H), 3.40 (dd, 1H, J_1 = 20.0 Hz, J_2 = 12.0 Hz), 2.80–2.77 (m, 1H); 13C NMR (100 MHz, CDCl3): δ 169.1, 152.9, 147.8, 138.1, 136.5, 136.3, 136.2, 134.2, 127.8, 127.3, 121.5, 121.3, 116.4, 103.7, 60.7, 55.9, 54.4, 39.2, 31.2; FT-IR (DCM): 3120, 1611, 1436, 1346, 771 cm[−]¹ ; HRMS (ESI): m/z [M + H]⁺ calcd for $C_{32}H_{35}N_2O_7$: 559.2444; found 559.2455.

(1S*,2S*,4R*)-2-(4-Methoxyphenyl)-N-(quinolin-8-yl)- 4-(3,4,5-trimethoxyphenyl)cyclobutanecarboxamide (16ai). Analytical TLC on silica gel, 2.5:2.5 ethyl acetate/hexanes R_f = 0.40 red color liquid; 50 mg, 40% yield; ¹H NMR (400 MHz, CDCl₃): δ 9.52 (br s, 1H), 8.73 (q, 1H, J = 4.0 Hz), 8.42 (q, 1H, J $= 8.0$ Hz), 8.10 (q, 1H, J = 4.0 Hz), 7.42–7.26 (m, 5H), 6.82 (q, $2H, J = 4.0 \text{ Hz}$, 6.50 (d, $2H, J = 8.0 \text{ Hz}$), 6.55 (t, $2H, J = 4.0 \text{ Hz}$), 4.15−4.10 (m, 1H), 4.02−3.97 (m, 2H), 3.73 (s, 3H), 3.71 (s, 6H), 3.61 (s, 3H), 3.44 (dd, 1H, $J_1 = 20.0$ Hz, $J_2 = 12.0$ Hz), 2.25−2.72 (m, 1H); ¹³C NMR (100 MHz, CDCl₃): δ 169.2, 157.8, 152.8, 147.8, 138.1, 136.4, 136.3, 136.2, 134.2, 132.7, 127.8, 127.7, 127.3, 121.4, 121.1116.4, 113.6, 103.8, 60.6, 55.8, 55.2, 54.6, 30.7, 38.1, 30.8; FT-IR (DCM): 2921, 1523, 1476, 1444, 749 cm[−]¹ ;HRMS (ESI): m/z [M + H]⁺ calcd for $C_{30}H_{31}N_2O_5$: 499.2233; found 499.2234.

(1R*,2R*,4S*)-2-(3,4-Dimethoxyphenyl)-N-(quinolin-8-yl)-4-(3,4,5-trimethoxyphenyl)cyclobutanecarboxamide (16aj). Analytical TLC on silica gel, 2.5:2.5 ethyl acetate/ hexanes R_f = 0.30 light-brown color liquid; 99 mg, 75% yield; 1 H NMR (400 MHz, CDCl₃): δ 9.57 (br s, 1H), 8.72 (q, 1H, J = 4.0 Hz), 8.44 (dd, 1H, $J_1 = 8.0$ Hz, $J_2 = 4.0$ Hz), 8.10 (q, 1H, $J = 4.0$ Hz), 7.41–7.32 (m, 3H), 6.90 (q, 1H, J = 4.0 Hz), 6.77 (t, 1H, J = 8.0 Hz), 6.55 (t, 2H, J = 4.0 Hz), 4.14–4.11 (m, 1H), 4.02–3.98 (m, 2H), 3.78 (s, 6H), 3.74 (s, 3H), 3.72 (s, 6H), 3.65 (s, 3H), 3.43 (dd, 1H, J₁ = 20.0 Hz, J₂ = 12.0 Hz), 2.77–2.76 (m, 1H); ¹³C NMR (100 MHz, CDCl₃): δ 169.2, 152.9, 148.6, 147.8, 147.3, 138.1, 136.6, 136.3, 136.1, 134.2, 133.3, 127.7, 127.3, 121.4, 121.2, 119.0, 116.4, 110.8, 110.4, 103.8, 60.7, 55.9, 55.6, 54.5, 39.4, 38.6, 31.3; FT-IR (DCM): 2811, 1535, 1476, 1423, 981 cm⁻¹; HRMS (ESI): m/z [M + H]⁺ calcd for C₃₁H₃₃N₂O₆: 529.2338; found 529.2343.

(1R*,2S*)-2-(4-Methoxyphenyl)-N-(quinolin-8-yl) cyclobutanecarboxamide (17a). Analytical TLC on silica gel, 1:4 ethyl acetate/hexane $R_f = 0.70$; brown color liquid; 25 mg, 30% yield; ¹H NMR (400 MHz, CDCl₃): δ 9.36 (br s, 1H), 8.73 $(q, 1H, J = 8.0 \text{ Hz})$, 8.54 $(q, 1H, J = 8.0 \text{ Hz})$, 8.11 $(q, 1H, J = 8.0 \text{ Hz})$ Hz), 7.47−7.39 (m, 3H), 7.25−7.23 (m, 2H), 6.65 (q, 2H, J = 4.0 Hz), 4.08 (dd, 1H, $J_1 = 20.0$ Hz, $J_2 = 8.0$ Hz), 3.75 (q, 1H, $J = 4.0$ Hz), 3.57 (s, 3H), 2.69−2.63 (m, 2H), 2.12−2.33 (m, 2H); 13C NMR (100 MHz, CDCl₃): δ 171.4, 158.0, 147.8, 138.2, 136.1, 134.4, 132.8, 128.4, 127.7, 127.3, 121.3, 120.9, 116.1, 113.5, 55.0, 47.8, 42.8, 25.4, 20.4, ; FT-IR (DCM): 3231, 1610, 1494, 1321, 710 cm⁻¹; HRMS (ESI): m/z [M + H]⁺ calcd for C₂₁H₂₁N₂O₂: 333.1603; found 333.1599.

(1R*,2S*)-2-Phenyl-N-(quinolin-8-yl)cyclobutanecarboxamide (17b). Analytical TLC on silica gel, 1:4 ethyl acetate/hexane $R_f = 0.70$; brown color liquid; 15 mg, 20% yield; ¹H NMR (400 MHz, CDCl₃): δ 9.40 (br s, 1H), 8.73 (t, 1H, J = 8.0 Hz), 8.54 (q, 1H, $I = 8.0$ Hz), 8.10 (q, 1H, $I = 8.0$ Hz), 7.46− 7.36 (m, 2H), 7.34−7.32 (m, 2H), 7.32−7.28 (m, 1H), 7.24− 7.09 (m, 2H), 6.98–6.94 (m, 1H), 4.12 (dd, 1H, $J_1 = 20.0$ Hz, $J_2 =$ 8.0 Hz), 3.80 (q, 1H, J = 4.0 Hz), 2.75−2.64 (m, 2H), 2.45−2.36 $(m, 2H);$ ¹³C NMR (100 MHz, CDCl₃): δ 171.3, 147.8, 140.7, 138.2, 136.1, 134.3, 128.0, 127.4, 127.3, 126.9, 126.3, 121.3, 121.1, 116.1, 47.7, 43.4, 25.1, 20.6; FT-IR (DCM): 3355, 1650, 1534, 1475, 890 cm[−]¹ ; HRMS (ESI): m/z [M + H]⁺ calcd for $C_{20}H_{19}N_2O: 303.1497$; found 303.1504.

(1R*,2S*)-N-(Quinolin-8-yl)-2-(thiophen-2-yl)cyclobutanecarboxamide (17c). Analytical TLC on silica gel, 1:4 ethyl acetate/hexane $R_f = 0.70$; yellow color liquid; 32 mg, 42% yield; ¹H NMR (400 MHz, CDCl₃): δ 9.51 (br s, 1H), 8.76 (q, 1H, $J = 8.0$ Hz), 8.65 (q, 1H, $J = 4.0$ Hz), 8.12 (q, 1H, $J = 8.0$ Hz), $7.49-7.40$ (m, 3H), 6.98–6.93 (m, 2H), 6.76 (q, 1H, J = 4.0 Hz), 4.35−4.35 (dd, 1H, J_1 = 20.0 Hz, J_2 = 8.0 Hz), 3.79 (q, 1H, J_2 = 4.0 Hz), 2.71−2.53 (m, 3H), 2.36−2.30 (m, 1H); 13C NMR (100 MHz, CDCl₃): δ 170.8, 147.9, 144.2, 138.3, 136.2, 134.4, 127.8, 127.4, 126.7, 124.7, 123.7, 121.4, 121.1, 116.2, 48.2, 38.7, 27.8, 20.6; FT-IR (KBr): 3335, 1641, 1510, 1431, 790 cm⁻¹; HRMS (ESI): m/z [M + H]⁺ calcd for C₁₈H₁₇N₂OS: 309.1061; found 309.1071.

(1R*,2S*)-N-(Quinolin-8-yl)-2-(3,4,5-trimethoxyphenyl)cyclobutanecarboxamide (17d). Analytical TLC on silica gel, 2.5:2.5 ethyl acetate/hexanes $R_f = 0.40$; white solid 31 mg, 32% yield; mp 125−127 °C; $^1\text{H NMR}$ (400 MHz, CDCl₃): δ 9.26 (br s, 1H), 8.67–8.62 (m, 2H), 8.08 (q, 1H, $J = 8.0$ Hz), 7.46−7.39 (m, 3H), 6.50 (s, 2H), 4.08−4.05 (m, 1H), 4.04−3.71 (m, 1H), 3.67 (s, 6H), 3.67 (s, 6H), 3.30 (s, 3H), 2.70−2.60 (m, 2H), 2.38–2.28 (m, 2H); ¹³C NMR (100 MHz, CDCl₃): δ 171.4, 152.8, 147.7, 138.1, 136.4, 136.3, 136.2, 134.5, 127.8, 127.2, 121.4, 121.1, 115.8, 104.3, 60.3, 55.8, 48.2, 43.9, 25.5, 19.9; FT-IR (KBr): 2830, 1599, 1502, 1391, 881 cm[−]¹ ; HRMS (ESI): m/z [M + H]⁺ calcd for C₂₃H₂₅N₂O₄: 393.1814; found 393.1808.

(1S*,2R*,4S*)-2,4-Bis(3-formylphenyl)-N-(2-(methylthio)phenyl)cyclobutanecarboxamide (19a). Analytical TLC on silica gel, 1.5:3.5 ethyl acetate/hexane $R_f = 0.30$; brown color solid; 56 mg, 52% yield; mp 100−102 °C; $^1\rm H$ NMR $(400 \text{ MHz}, \text{CDCl}_3): \delta 10.1 \text{ (s, 2H)}, 7.95 \text{ (br s, 1H)}, 7.84 \text{ (s, 2H)},$ 7.69 (d, 2H, J = 8.0 Hz), 7.61 (d, 2H, J = 8.0 Hz), 7.50−7.44 (m, 1H), 7.29 (q, 2H, J = 4.0 Hz), 7.02 (q, 1H, J = 8.0 Hz), 6.98–6.90

 $(m, 2H)$, 4.17–4.04 $(m, 3H)$, 3.64 (dd, 1H, $J_1 = 20.0$ Hz, $J_2 = 8.0$ Hz), 2.82−2.79 (m, 1H), 2.11 (s, 3H); 13C NMR (100 MHz, CDCl3): δ 192.5, 168.0, 141.4, 137.1, 136.3, 133.1, 132.1, 128.9, 128.3, 125.6, 124.5, 121.0, 54.1, 38.5, 29.5, 18.5; FT- IR (KBr): 3351, 1668, 1582, 1398, 810 cm⁻¹; HRMS (ESI): m/z [M + H]⁺ calcd for $C_{26}H_{24}NO_3S$: 430.1476; found 430.1482.

(1S*,2R*,4S*)-N-(2-(Methylthio)phenyl)-2,4-di-p-tolylcyclobutanecarboxamide (19b). Analytical TLC on silica gel, 1:4 ethyl acetate/hexane $R_f = 0.70$; white color solid; 45 mg, 45% yield; mp 98−100 °C; ¹H NMR (400 MHz, CDCl₃): δ 7.95 (br s, 1H),7.74 (d, 1H, J = 8.0 Hz), 7.34 (q, 1H, J = 8.0 Hz), 7.29–7.21 (m, 4H), 7.09−7.05 (m, 5H), 6.94−6.91 (m, 1H), 4.01−3.94 (m, 3H), 3.48 (dd, 1H, J₁ = 12.0 Hz, J₂ = 8.0 Hz), 2.70–2.67 (m, 1H), 2.28 (s, 6H), 2.14 (s, 3H); ¹³C NMR (100 MHz, CDCl₃): δ 168.8, 138.2, 137.3, 135.6, 132.8, 129.0, 128.9, 128.6, 126.8, 123.8, 121.0, 54.5, 38.7, 30.0, 21.1, 18.8; FT-IR (KBr): 3324, 1622, 1534, 1491, 810 cm[−]¹ ; HRMS (ESI): m/z [M + H]⁺ calcd for $C_{26}H_{28}NOS: 402.1891$; found 402.1896.

(1S*,2R*,4S*)-2,4-Bis(4-acetylphenyl)-N-(2-(methylthio)phenyl)cyclobutanecarboxamide (19c). Analytical TLC on silica gel, 2:3 ethyl acetate/hexane $R_f = 0.60$; white color solid; 33 mg, 29% yield; mp 189–191 °C; ¹H NMR (400 MHz, CDCl₃): δ 7.99 (br s, 1H), 7.94 (q, 4H, J = 8.0 Hz), 7.61 (q, 1H, $J = 8.0$ Hz), 7.39 (q, 4H, $J = 8.0$ Hz), 7.33–7.28 (m, 1H), 7.05−7.03 (m, 1H), 7.01−6.93 (m, 1H), 4.11−4.06 (m, 3H), 3.57 (dd, 1H, J₁ = 12.0 Hz, J₂ = 8.0 Hz), 2.77–2.67 (m, 1H), 2.56 $(s, 6H)$, 2.30 $(s, 3H)$; ¹³C NMR (100 MHz, CDCl₃): δ 197.9, 168.0, 146.0, 137.3, 135.3, 134.9, 132.3, 128.5, 128.4, 128.3, 127.0, 125.3, 124.4, 120.9, 54.3, 38.9, 29.7, 26.6, 18.3; FT-IR (KBr): 3350, 1620, 1551, 1441, 710 cm[−]¹ ; HRMS (ESI): m/z [M $+$ Na]⁺ calcd for C₂₈H₂₇NO₃SNa: 480.1609; found 480.1604.

(1S*,2R*,4S*)-N-(2-(Dimethylamino)ethyl)-2,4-bis(3 nitrophenyl)cyclobutanecarboxamide (21a). Analytical TLC on silica gel, 2:3.methanol/ethyl acetate $R_f = 0.40$; black color solid; 46 mg, 45% yield; mp 130–132 °C; ¹H NMR (400 MHz, CDCl₃): δ 8.07–8.01 (m, 4H), 7.60 (d, 2H, J = 12.0 Hz), 7.46 (t, 2H, J = 8.0 Hz), 6.34 (br s, 1H), 4.03–3.89 (m, 2H), 3.87 $(q, 1H, J = 4.0 \text{ Hz})$, 3.51 (dd, 1H, $J_1 = 20.0 \text{ Hz}, J_2 = 12.0 \text{ Hz}$), 2.83−2.72 (m, 3H), 2.01 (s, 6H), 1.96 (t, 2H, J = 4.0 Hz); ¹³C NMR (100 MHz, CDCl₃): δ 169.0, 148.0, 143.0, 133.2, 129.0, 121.9, 121.3, 57.4, 52.5, 44.7, 37.9, 36.0, 29.7; FT-IR (KBr): 2876, 2325, 1618, 1532, 817 cm⁻¹; HRMS (ESI): *m*/z [M + H]⁺ calcd for $C_{21}H_{25}N_4O_5$: 413.1824; found 413.1813.

(1S*,2R*,4S*)-N-(2-(Dimethylamino)ethyl)-2,4-diphenylcyclobutanecarboxamide (21b). Analytical TLC on silica gel, 2:3. methanol/ethyl acetate $R_f = 0.40$; red color liquid; 24 mg, 30% yield; ¹H NMR (400 MHz, CDCl₃): δ 7.33–7.17 (m, 10H), 5.78 (br s, 1H), 3.96–3.89 (m, 2H), 3.78 (q, 1H, $J = 4.0$ Hz), 3.41 (dd, 1H, $J_1 = 20.0$ Hz, $J_2 = 12.0$ Hz), 2.83–2.78 (q, 2H, $J = 4.0$ Hz), 2.65−2.61(q, 1H, $J = 4.0$ Hz), 2.0 (s, 6H), 1.86 (t, 2H, J = 4.0 Hz); ¹³C NMR (100 MHz, CDCl₃): δ 169.9, 140.9, 127.9, 127.1, 126.1, 57.4, 53.3, 44.8, 38.6, 35.9, 29.5; FT-IR (DCM): 2912, 2324, 1612, 1543, 809 cm[−]¹ ; HRMS (ESI): m/z $[M + H]^{+}$ calcd for $C_{21}H_{27}N_{2}O: 323.2123$; found 323.2113.

(1S*,2R*,4S*)-N-(2-(Dimethylamino)ethyl)-2,4-bis(4 methoxyphenyl)cyclobutanecarboxamide (21c). Analytical TLC on silica gel, 2:3. methanol/ethyl acetate $R_f = 0.40$; paleyellow color liquid; 51 mg, 53% yield; ¹H NMR (400 MHz, CDCl₃): δ 7.26−7.17 (m, 4H), 6.84−6.80 (m, 4H), 5.67 (br s, 1H), 3.84−3.79 (m, 2H), 3.75 (s, 6H), 3.66−3.63 (m, 1H),3.26 (dd, 1H, J₁ = 20.0 Hz, J₂ = 12.0 Hz), 2.84−2.79 (m, 2H), 2.56− 2.52 (m, 1H), 1.98 (s, 6H), 1.85 (t, 2H, $J = 4.0$ Hz); ¹³C NMR $(100 \text{ MHz}, \text{CDCl}_3)$: δ 170.2, 157.9, 132.9, 128.2, 113.4, 57.6, 55.2, 53.5, 44.8, 38.0, 36.0, 30.1; FT-IR (DCM,): 2948, 2213, 1623, 1534, 812 cm[−]¹ ; HRMS (ESI): m/z [M + H]⁺ calcd for $C_{23}H_{31}N_2O_3$: 383.2335; found 383.2352.

(1S*,2R*,4S*)-N-(2-(Dimethylamino)ethyl)-2,4-di-ptolylcyclobutanecarboxamide (21d). Analytical TLC on silica gel, 2:3. methanol/ethyl acetate $R_f = 0.40$; brown color liquid; 35 mg, 40% yield; ¹H NMR (400 MHz, CDCl₃): δ 7.19− 7.11 (m, 8H), 5.70 (br s, 1H), 3.89−3.84 (m, 2H), 3.72 (q, 1H, J $= 4.0$ Hz), 3.32 (dd, 1H, $J_1 = 20.0$ Hz, $J_2 = 12.0$ Hz), 2.86–2.82 $(m, 2H)$, 2.61 (q, 1H, J = 4.0 Hz), 2.33 (s, 6H), 2.01 (s, 6H), 1.85 $(t, 2H, J = 4.0 \text{ Hz})$; ¹³C NMR (100 MHz, CDCl₃): δ 170.1, 137.8, 135.4, 128.6, 127.2, 57.5, 53.4, 44.8, 38.4, 36.0, 29.7, 21.1; FT-IR (DCM): 2900, 2316, 1627, 1553, 737 cm[−]¹ ; HRMS (ESI): m/z $[M + H]^{+}$ calcd for $C_{23}H_{31}N_{2}O: 351.2436$; found 351.2429.

(1S*,2R*,4S*)-N-(2-(Dimethylamino)ethyl)-2,4-bis(4 ethylphenyl)cyclobutanecarboxamide (21e). Analytical TLC on silica gel, 2:3. methanol/ethyl acetate $R_f = 0.40$; brown color solid; 37 mg, 39% yield; mp 185−188 °C; ¹H NMR $(400 \text{ MHz}, \text{CDCl}_3)$: δ 7.21–7.12 (m, 8H), 6.50 (br s, 1H), 3.88 $(q, 2H, J = 4.0 \text{ Hz})$, 3.73 $(q, 1H, J = 4.0 \text{ Hz})$, 3.33 $(dd, 1H, J_1 =$ 20.0 Hz, J_2 = 12.0 Hz), 2.91–2.87 (m, 3H), 2.61 (q, 4H, J = 4.0 Hz), 2.05 (s, 6H), 1.94 (t, 2H, $J = 4.0$ Hz), 1.24 (t, 6H, $J = 4.0$ Hz); ¹³C NMR (100 MHz, CDCl₃): δ 170.4, 141.8, 138.1, 127.4, 127.1, 57.4, 53.2, 44.1, 38.4, 35.6, 29.7, 28.5, 15.7; FT-IR (DCM): 3240, 1634, 1494, 1344, 710 cm[−]¹ ; HRMS (ESI): m/z $[M + H]^{+}$ calcd for $C_{25}H_{35}N_{2}O$: 379.2749; found 3792746.

(1S*,2R*,4S*)-2,4-Bis(4-chlorophenyl)-N-(2-(dimethylamino)ethyl)cyclobutanecarboxamide (21f). Analytical TLC on silica gel, 2:3. methanol/ethyl acetate $R_f = 0.40$; white color solid; 40 mg, 41% yield; mp 155−157 °C; ¹H NMR (400 MHz, CDCl₃): δ 7.28–7.18 (m, 8H), 5.71 (br s, 1H), 3.88–3.81 $(m, 2H)$, 3.70–3.65 $(m, 1H)$, 3.36 (dd, 1H, J₁ = 20.0 Hz, J₂ = 12.0 Hz), 2.86−2.82 (m, 2H), 2.62−2.55 (m, 1H), 1.90 (s, 6H), 1.89 $(t, 2H, J = 4.0 \text{ Hz})$; ¹³C NMR (100 MHz, CDCl₃): δ 169.4, 139.1, 131.9, 128.5, 128.1, 57.5, 53.2, 44.9, 38.1, 36.0, 29.7; FT-IR (KBr): 2912, 2323, 1618, 1512, 812 cm[−]¹ ; HRMS (ESI):m/z [M + H]⁺ calcd for $C_{21}H_{25}Cl_2N_2O$: 391.1343; found 391.1340.

(1S*,2R*,4S*)-2,4-Bis(4-bromophenyl)-N-(2-(dimethylamino)ethyl)cyclobutanecarboxamide (21g). Analytical TLC on silica gel, 2:3. methanol/ethyl acetate $R_f = 0.40$; brown color liquid; 54 mg, 45% yield; ¹ H NMR (400 MHz, CDCl₃): δ 7.41–7.38 (m, 4H), 7.14 (d, 4H, J = 8.0 Hz), 6.50 (br s, 1H), 3.82−3.77 (m, 2H), 3.70 (q, 1H, J = 4.0 Hz), 3.33 (dd, $1H, J_1 = 20.0$ Hz, $J_2 = 12.0$ Hz), 2.92–2.82 (m, 2H), 2.57 (t, 1H, J $= 4.0$ Hz), 2.09 (s, 6H), 2.0 (t, 2H, J = 4.0 Hz); ¹³C NMR (100 MHz, CDCl₃): δ 169.6, 139.8, 130.9, 129.1, 119.9, 57.5, 52.9, 44.4, 38.1, 35.5, 29.5; FT-IR (KBr): 2912, 2316, 1637, 1544, 917 cm⁻¹ HRMS (ESI): m/z [M + H]⁺ calcd for C₂₁H₂₅Br₂N₂O: 479.0333; found 479.0316.

(1S*,2R*,4S*)-N-(2-(Dimethylamino)ethyl)-2,4-bis(4 nitrophenyl)cyclobutanecarboxamide (21h). Analytical TLC on silica gel, 2.5:2.5. methanol/ethyl acetate $R_f = 0.40$; yellow color liquid; 44 mg, 43% yield; ¹H NMR (400 MHz, CDCl₃ and DMSO- d_6): δ 8.08 (d, 4H, J = 8.0 Hz), 7.36–7.28 (m, 4H), 6.56 (br s, 1H), 3.96−3.91 (m, 2H), 3.86 (q, 1H, J = 4.0 Hz), 3.42 (dd, 1H, $J_1 = 20.0$ Hz, $J_2 = 12.0$ Hz), 2.81–2.77 (m, 2H), 2.68 (q, 2H, $J = 4.0$ Hz), 2.0 (s, 6H), 1.92 (t, 2H, $J = 8.0$ Hz); ¹³C NMR (100 MHz, CDCl₃ and DMSO- d_6): δ 169.0, 148.8, 146.3, 127.7, 123.2, 57.6, 53.0, 44.7, 38.2, 35.9, 29.5; FT-IR (KBr): 3243, 1632, 1521, 1494, 810 cm[−]¹ ; HRMS (ESI):m/z [M $+ H$]⁺ calcd for C₂₁H₂₅N₄O₅: 413.1824; found 413.1823.

(1S*,2R*,4S*)-2,4-Bis(4-cyanophenyl)-N-(2-(dimethylamino)ethyl)cyclobutanecarboxamide (21i). Analytical TLC on silica gel, 2.5:2.5. methanol/ethyl acetate $R_f = 0.40$; black color solid; 45 mg, 48% yield; mp 159−161 °C; ¹ H NMR $(400 \text{ MHz}, \text{CDCl}_3): \delta 7.58 - 7.55 \text{ (m, 4H)}, 7.35 - 7.28 \text{ (m, 2H)},$ 6.13 (br s, 1H), 3.97–3.90 (m, 2H), 3.82 (q, 1H, J = 4.0 Hz), 3.45 (dd, 1H, $J_1 = 20.0$ Hz, $J_2 = 12.0$ Hz), 2.83 (q, 2H, $J = 4.0$ Hz), 2.66−2.63 (m, 1H), 2.03 (s, 6H), 1.94 (t, 2H, $J = 8.0$ Hz); ¹³C NMR (100 MHz, CDCl₃): δ 168.9, 146.4, 131.8, 127.8, 119.1, 109.8, 57.5, 53.0, 45.1, 44.8, 38.4, 35.9, 29.2; FT- IR (KBr): 2945, 2226, 1607, 1506, 827 cm⁻¹; HRMS (ESI): *m*/z [M + H]⁺ calcd for $C_{23}H_{25}N_{4}O: 373.2028$; found 373.2026.

(1S*,2R*,4S*)-2,4-Bis(4-bromo-3-fluorophenyl)-N-(2- (dimethylamino)ethyl)cyclobutanecarboxamide (21j). Analytical TLC on silica gel, 2.5:2.5. methanol/ethyl acetate R_f = 0.40; red color liquid; 68 mg, 53% yield; 1 H NMR (400 MHz, CDCl₃): δ 7.46–7.42 (m, 2H), 7.03 (q, 2H, J = 4.0 Hz), 6.92 (q, 2H, J = 8.0 Hz), 6.45 (br s, 1H), 3.82−3.71 (m, 2H), 3.70 (q, 1H, $J = 4.0 \text{ Hz}$), 3.23 (dd, 1H, $J_1 = 20.0 \text{ Hz}$, $J_2 = 12.0 \text{ Hz}$), 2.94–2.90 $(m, 2H)$, 2.61−2.58 $(m, 1H)$, 2.10 $(s, 6H)$, 2.03 $(t, 2H, J = 4.0)$ Hz); ¹³C NMR (100 MHz, CDCl₃): δ 169.0, 160.1, 157.5 (d, J_{C-F} = 246 Hz), 142.6 (d, J_{C-F} = 7.0 Hz), 132.9, 123.9 (d, J_{C-F} = 3.0 Hz), 115.2 (d, $J_{C-F} = 22.0$ Hz), 106.3 (d, $J_{C-F} = 20.0$ Hz), 57.6, 52.8, 44.8, 37.8, 36.0, 29.7; FT-IR (DCM): 3252, 1601, 1511, 1468, 710 cm[−]¹ ; HRMS (ESI): m/z [M + H]+ calcd for $C_{21}H_{23}Br_2F_2N_2O$: 515.0145; found 515.0101.

(1S*,2R*,4S*)-N-(2-(Dimethylamino)ethyl)-2,4-bis(3,4 dimethylphenyl)cyclobutanecarboxamide (21k). Analytical TLC on silica gel, 2:3. methanol/ethyl acetate $R_f = 0.40$ black color solid; 40 mg, 42% yield; mp 220–222 °C; ¹H NMR (400 MHz, CDCl₃): δ 7.08–7.01 (m, 6H), 6.08 (br s, 1H), 3.85–3.81 $(m, 2H)$, 3.71 (t, 1H, J = 4.0 Hz), 3.32 (dd, 1H, J₁ = 20.0 Hz, J₂ = 12.0 Hz), 2.92−2.88 (m, 2H), 2.58−2.56 (m, 1H), 2.27 (s, 6H), 2.24 (s, 6H), 2.04 (s, 6H), 1.91 (t, 2H, $J = 4.0$ Hz); ¹³C NMR $(100 \text{ MHz}, \text{CDCl}_3)$: δ 170.4, 138.3, 135.9, 134.0, 129.3, 128.4, 124.5, 57.5, 53.2, 44.5, 38.4, 35.8, 29.7, 19.9, 19.4; FT-IR (KBr): 2937, 2316, 1567, 1516, 918 cm⁻¹; HRMS (ESI): *m*/z [M + H]⁺ calcd for $C_{25}H_{35}N_2O: 379.2749$; found 379.2748.

(1S*,2R*,4S*)-2,4-Bis(2,3-dihydrobenzo[b][1,4]dioxin-5-yl)-N-(2-(dimethylamino)ethyl)cyclobutanecarboxamide (21l). Analytical TLC on silica gel, 2:3. methanol/ethyl acetate R_f = 0.40 brown color solid; 45 mg, 41% yield; ¹H NMR $(400 \text{ MHz}, \text{CDCl}_3): \delta 6.77 - 6.70 \text{ (m, 6H)}, 6.11 \text{ (br s, 1H)}, 4.21$ $(q, 8H, J = 4.0 \text{ Hz})$, 3.77–3.70 (m, 2H), 3.64 (q, 1H, J = 4.0 Hz), 3.15 (dd, 1H, $J_1 = 20.0$ Hz, $J_2 = 12.0$ Hz), 2.94 (q, 2H, $J = 4.0$ Hz), 2.52−2.49 (m, 1H), 2.09 (s, 6H), 2.10 (t, 2H, J = 4.0 Hz); ¹³C NMR (100 MHz, CDCl₃): δ 170.2, 143.0, 141.8, 134.3, 120.0, 116.7, 115.9, 64.3, 57.6, 52.9, 44.8, 44.7, 37.8, 35.9, 30.0; FT-IR (DCM): 3303, 1651, 1588, 1425, 890 cm[−]¹ ; HRMS (ESI): m/z $[M + H]^{+}$ calcd for $C_{25}H_{31}N_{2}O_{5}$: 439.2223; found 439.2233.

(1S*,2R*,4S*)-N-(2-(Dimethylamino)ethyl)-2,4-di- (thiophen-2-yl)cyclobutanecarboxamide (21m). Analytical TLC on silica gel, 2:3. methanol/ethyl acetate $R_f = 0.40$ black color liquid; 25 mg, 30% yield; ¹H NMR (400 MHz, CDCl₃): δ 7.17 (q, 2H, J = 4.0 Hz), 6.99–6.94 (m, 4H), 6.05 (br s, 1H), 4.03−3.98 (m, 2H), 3.61(q, 1H, J = 4.0 Hz), 3.35 (dd, 1H, J₁ = 20.0 Hz, $J_2 = 12.0$ Hz), 3.02 (q, 2H, $J = 4.0$ Hz), 2.80 (q, 1H, $J =$ 4.0 Hz), 2.07 (s, 6H), 2.03 (t, 2H, J = 4.0 Hz); ¹³C NMR (100 MHz, CDCl₃): δ 169.3, 143.7, 126.7, 125.1, 123.8, 57.6, 54.7, 48.1, 36.1, 35.4, 35.0; FT-IR (DCM): 2800, 2312, 1621, 1512, 800 cm⁻¹; HRMS (ESI): m/z [M + H]⁺ calcd for C₁₇H₂₃N₂OS₂: 335.1251; found 335.1245.

N-(((1S*,2R*,4S*)-2,4-Bis(4-bromophenyl)cyclobutyl) methyl)quinolin-8-amine (23). Analytical TLC on silica gel, 1:4 ethyl acetate/hexanes R_f . = 0.70 brown color liquid; 51 mg,

40% yield; ¹H NMR (400 MHz, CDCl₃): δ 8.65 (q, 1H, J = 4.0) Hz), 8.07 (q, 1H, J = 4.0 Hz), 7.50–7.17 (m, 10H), 7.07–7.04 $(m, 1H)$, 6.62–6.59 $(m, 1H)$, 3.59 $(q, 2H, J = 4.0 Hz)$, 3.38–3.26 (m, 2H), 2.77−2.74 (m, 2H), 2.22−2.13 (m, 1H); 13C NMR $(100 \text{ MHz}, \text{CDCl}_3)$: δ 146.8, 142.6, 138.1, 135.9, 131.5, 131.4, 128.5, 127.6, 127.5, 126.9, 120.1, 114.0, 104.7, 51.0, 46.8, 40.3, 34.0; FT-IR (DCM): 3234, 1644, 1412, 1382, 819 cm⁻¹; HRMS (ESI): m/z [M + H]⁺ calcd for C₂₆H₂₃Br₂N₂: 521.0228; found 521.0229.

(1R*,2R*,4S*)-2,4-Di-p-tolylcyclobutanecarboxylic acid (24). Following the general procedure described above, 24 was obtained as a brown color liquid (crude material was almost pure); 67 mg, 96% yield; ¹H NMR (400 MHz, CDCl₃): δ 7.36− 7.17 (m, 8H), 3.79 (q, 2H, $J = 8.0$ Hz), 3.30 (t, 1H, $J = 12.0$ Hz), 2.79−2.76 (m, 1H), 2.38 (s, 6H), 2.37−2.27 (m, 1H); 13C NMR $(100 \text{ MHz}, \text{CDCl}_3)$: δ 179.7, 139.7, 136.3, 129.2, 126.5, 52.6, 39.2, 32.9, 21.1; FT-IR (DCM): 2833, 1515, 1485, 1354, 806 cm^{-1} ; HRMS (ESI): m/z [M - H] $\frac{1}{2}$ calcd for $\text{C}_{19}\text{H}_{19}\text{O}_2$: 279.1385; found 279.1389.

(1R*,2R*,4S*)-2,4-Bis(4-chlorophenyl)cyclobutanecarboxylic acid (25). Following the general procedure described above, 25 was obtained as a white color liquid (crude material was almost pure); 78 mg, 98% yield; ¹H NMR $(400 \text{ MHz}, \text{CDCl}_3)$: δ 7.35–7.32 (m, 8H), 3.80 (q, 2H, J = 8.0 Hz), 3.27 (t, 1H, J = 12.0 Hz), 2.84–2.78 (m, 1H), 2.31 (g, 1H, J $= 8.0$ Hz); ¹³C NMR (100 MHz, CDCl₃): δ 179.6, 140.8, 132.6, 128.6, 128.0, 52.3, 38.8, 32.5; FT-IR (DCM): 2966, 1565, 1432, 1331, 800 cm⁻¹; HRMS (ESI): m/z [M - H][−] calcd for $C_{17}H_{13}Cl_2O_2$: 319.0292; found 319.0298.

(1R*,2R*,4S*)-2,4-Bis(4-bromophenyl)cyclobutanecarboxylic acid (26). Following the general procedure described above, 26 was obtained as a brown color solid (crude material was almost pure); 97 mg, 95% yield; mp131− 133 °C; ¹H NMR (400 MHz, CDCl₃): δ 7.50−7.46 (m, 4H), 7.21−7.19 (m, 4H), 3.79 (q, 2H, J = 8.0 Hz), 3.27 (t, 1H, J = 12.0 Hz), 2.84−2.77 (m, 1H), 2.30−2.15 (m, 1H); 13C NMR (100 MHz, CDCl₃): δ 179.5, 141.3, 131.6, 128.3, 120.6, 52.2, 38.8, 32.3; FT-IR (KBr): 2922, 1698, 1516, 1425, 806 cm⁻¹; HRMS (ESI): m/z [M + Na]⁺ calcd for C₁₇H₁₄Br₂O₂Na: 430.9258; found 430.9281.

(1S*,2R*,4S*)-2,4-Bis(4-bromophenyl)-N-methyl-N- (quinolin-8-yl)cyclobutanecarboxamide (27). Analytical TLC on silica gel, 1:4 ethyl acetate/hexanes $R_f = 0.70$ lightyellow color liquid; 130 mg, 95% yield; ¹H NMR (400 MHz, CDCl₃): δ 8.98 (q, 1H, J = 4.0 Hz), 8.26 (q, 1H, J = 4.0 Hz), 7.85 $(q, 1H, J = 8.0 \text{ Hz})$, 7.54–7.46 (m, 6H), 7.26–7.23 (m, 4H), 6.72 $(q, 1H, J = 4.0 \text{ Hz})$, 3.54–3.36 (m, 4H), 2.98 (s, 3H), 2.48 (q, 1H, $J = 4.0$ Hz); ¹³C NMR (100 MHz, CDCl₃): δ 170.2, 151.0, 144.2, 141.4, 141.3, 139.2, 136.3, 131.0, 130.8, 130.4, 129.4, 128.8, 128.9, 128.1, 126.4, 122.1, 120.8, 119.2, 50.0, 39.7, 37.7, 37.1; FT-IR (DCM): 2941, 1613, 1523, 1468, 791 cm[−]¹ ; HRMS (ESI): m/z [M + H]⁺ calcd for C₂₇H₂₃Br₂N₂O: 549.0177; found 549.0175.

■ ASSOCIATED CONTENT

S Supporting Information

Copies of NMR spectra of all compounds and X-ray structures (ORTEP diagrams of the compounds 16c, 16f, 16g, 16m, 21a, 21f, 25 and 27) and crystal data. This material is available free of charge via the Internet at http://pubs.acs.org.

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Notes

The aut[hors declare no competing](mailto:sababu@iisermohali.ac.in) financial interest.

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■ REFERENCES

(1) (a) Dembisky, V. M. J. Nat. Med. 2008, 62, 1 and references cited therein. (b) For selected reviews on the chemistry of cyclobutanes, see the following: Namyslo, J. C.; Kaufmann, D. E. Chem. Rev. 2003, 103, 1485. (c) Demuth, M.; Mikhail, G. Synthesis 1989, 145. (d) Bach, T. Synthesis 1998, 683. (e) Lee-Ruff, E.; Madenova, G. Chem. Rev. 2003, 103, 1449. (f) Iriondo-Alberdi, J.; Greaney, M. F. Eur. J. Org. Chem. 2007, 4801. (g) Hoffmann, N. Chem. Rev. 2008, 108, 1052. (h) Rappoport, Z., Liebman, J. F., Eds. The Chemistry of Cyclobutanes; Wiley: Chichester, 2005.

(2) (a) Zhou, M.; Zhang, H.-B.; Wang, W.-G.; Gong, N.-B.; Zhan, R.; Li, X.-N.; Du, X.; Li, L.-M.; Li, Y.; Lu, Y.; Pu, J.-X.; Sun, H.-D. Org. Lett. 2013, 15, 4446. (b) Dai, J.; Jiménez, J. I.; Kelly, M.; Williams, P. G. J. Org. Chem. 2010, 75, 2399. (c) Lee, F.-P.; Chen, Y.-C.; Chen, J.-J.; Tsai, I.-L.; Chen, I.-S. Helv. Chim. Acta 2004, 87, 463. (d) Filho, R. B.; De Souza, M. P.; Mattos, M. E. O. Phytochemistry 1981, 20, 345. (e) Tsai, I.-L.; Lee, F.- P.; Wu, C.-C.; Duh, C.-Y.; Ishikawa, T.; Chen, J.-J.; Chen, Y.-C.; Seki, H.; Chen, I.-S. Planta Med. 2005, 71, 535. (f) Hill, R. A.; Sutherland, A. Nat. Prod. Rep. 2010, 27, 805.

(3) (a) Fujiwara, Y.; Naithou, K.; Miyazaki, T.; Morib Hashimoto, K.; Mori, K.; Yamamoto, Y. Tetrahedron Lett. 2001, 42, 2497. (b) Bucholtz, K. M.; Gareiss, P. C.; Tajc, S. G.; Miller, B. L. Org. Biomol. Chem. 2006, 4, 3973. (c) Tsukamoto, S.; Tomise, K.; Miyakawa, K.; Cha, B.-C.; Abe, T.; Hirota, H.; Ohta, T. Bioorg. Med. Chem. 2002, 10, 2981. (d) Martin, M. J.; Fernandez, R.; Francesch, A.; Amade, P.; de Matos-Pita, S. S.; Reyes, F.; Cuevas, C. Org. Lett. 2010, 12, 912. (e) Yang, C. S.; Wang, X. B.; Wang, J. S.; Luo, J. G.; Luo, J.; Kong, L. Y. Org. Lett. 2011, 13, 3380. (f) Marrero, J.; Rodriguez, A. D.; Baran, P.; Raptis, R. G.; Sanchez, J. A.; Ortega-Barria, E.; Capson, T. L. Org. Lett. 2004, 6, 1661.

(4) Chi, Y.-M.; Yan, W.-M.; Li, J.-S. Phytochemistry 1990, 29, 2376.

(5) (a) Tanaka, N.; Okasaka, M.; Ishimaru, Y.; Takaishi, Y.; Sato, M.; Okamoto, M.; Oshikawa, T.; Ahmed, S. U.; Consentino, L. M.; Lee, H. K.; Lee, H. K. Org. Lett. 2005, 7, 2997. (b) Nicolaou, K. C.; Sarlah, D.; Shaw, D. M. Angew. Chem., Int. Ed. 2007, 46, 4708. (c) Nicolaou, K. C.; Wu, T. R.; Sarlah, D.; Shaw, D. M.; Rowcliffe, E.; Burton, D. R. J. Am. Chem. Soc. 2008, 130, 11114. (d) Nicolaou, K. C.; Sanchini, S.; Sarlah, D.; Lu, G.; Wu, T. R.; Nomura, D. K.; Cravatt, B. F.; Cubitt, B.; de la Torre, J. C.; Hessell, A. J.; Burton, D. R. Proc. Natl. Acad. Sci. U.S.A. 2011, 108, 6715.

(6) (a) Li, Y. S.; Matsunaga, K.; Ishibashi, M.; Ohizumi, Y. J. Org. Chem. 2001, 66, 2165. (b) Mangion, I. K.; MacMillan, D. W. C. J. Am. Chem. Soc. 2005, 127, 3696.

(7) (a) Wei, K.; Li, W.; Koike, K.; Chen, Y. J.; Nikaido, T. J. Org. Chem. 2005, 70, 1164. (b) Matsuda, H.; Ninomiya, K.; Morikawa, T.; Yasuda, D.; Yamaguchi, I.; Yoshikawa, M. Bioorg. Med. Chem. 2009, 17, 7313.

(8) Takahashi, M.; Ichikawa, M.; Aoyagi, S.; Kibayashi, C. Tetrahedron Lett. 2005, 46, 57.

(9) Ichikawa, M.; Takahashi, M.; Aoyagi, S.; Kibayashi, C. J. Am. Chem. Soc. 2004, 126, 16553.

(10) Zhang, F.; Jia, Y. Tetrahedron 2009, 65, 6840.

(11) O'Malley, D. P.; Li, K.; Maue, M.; Zografos, A. L.; Baran, P. S. J. Am. Chem. Soc. 2007, 129, 4762.

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(12) Frébault, F.; Luparia, M.; Oliveira, M. T.; Goddard, R.; Maulide, N. Angew. Chem., Int. Ed. 2010, 49, 5672.

(13) Liu, R.; Zhang, M.; Wyche, T. P.; Winston-McPherson, G. N.; Bugni, T. S.; Tang, W. Angew. Chem., Int. Ed. 2012, 51, 7503.

(15) Tsai, A. S.; Bergman, R. G.; Ellman, J. A. J. Am. Chem. Soc. 2008, 130, 6316.

(16) Gutekunst, W. R.; Baran, P. S. J. Am. Chem. Soc. 2011, 133, 19076. (17) Gutekunst, W. R.; Baran, P. S. Angew. Chem., Int. Ed. 2012, 51, 7507.

(18) (a) Morrison, W. H., III; Hartley, R. D.; Himmelsbach, D. S. J. Agric. Food Chem. 1992, 40, 768. (b) Natarajan, A.; Ramamurthy, V. In The Chemistry of Cyclobutanes; Rappoport, Z., Liebman, J. F., Eds.; Wiley: Chichester, 2005; Vol. 1, p 807.

(19) (a) Cohen, M. D.; Schmidt, G. M. J. J. Chem. Soc. 1964, 2000. (b) Ramamurthy, V.; Venkatesan, K. Chem. Rev. 1987, 87, 433. (c) Ciamician, G.; Silber, P. Chem. Ber. 1908, 41, 1928. (d) Navarro, R.; Reisman, S. E. Org. Lett. 2012, 14, 4354. (e) Albrecht, L.; Dickmeiss, G.; Acosta, F. C.; Rodriguez-Escrich, C .; Davis, R. L.; Jorgensen, K. A. J. Am. Chem. Soc. 2012, 134, 2543. (f) Deng, J.; Hsung, R. P.; Ko, C. H. Org. Lett. **2012**, 14, 5562. (g) García-Expósito, E.; Bearpark, M. J.; Ortuño, R. M.; Robb, M. A.; Branchadell, V. J. Org. Chem. 2002, 67, 6070. (h) Araki, T.; Ozawa, T.; Yokoe, H.; Kanematsu, M.; Yoshida, M.; Shishido, K.Org. Lett. 2013, 15, 200.

(20) Lewis, F. D.; Quillen, S. L.; Hale, P. D.; Oxman, J. D. J. Am. Chem. Soc. 1988, 110, 1261.

(21) Fedorova, O.; Federov, Y. V.; Gulakova, E.; Schepel, N.; Alfimov, M.; Goli, U.; Saltiel, J. Photochem. Photobiol. Sci. 2007, 6, 1097.

(22) (a) Du, J.; Yoon, T. P. J. Am. Chem. Soc. 2009, 131, 14604. (b) Sqárez-Pantiga, S.; Hernandez-Diaz, C.; Rubio, E.; Gonzalez, J. M. Angew. Chem., Int. Ed. 2012, 51, 11552. (c) Luparia, M.; Oliveira, M. T.; Audisio, D.; Frébault, F.; Goddard, R.; Maulide, N. Angew. Chem., Int. Ed. 2011, 50, 12631.

(23) (a) Campeau, L. C.; Fagnou, K. Chem. Commun. 2006, 1253. (b) Alberico, D.; Scott, M. E.; Lautens, M. Chem. Rev. 2007, 107, 174. (c) Li, B.-J.; Yang, S.-D.; Shi, Z.-J. Synlett 2008, 949. (d) Ackermann, L.; Vicente, R.; Kapdi, A. R. Angew. Chem., Int. Ed. 2009, 48, 9792. (e) Yeung, C. S.; Dong, V. M. Chem. Rev. 2011, 111, 1215. (f) Liu, C.; Zhang, H.; Shi, W.; Lei, A. Chem. Rev. 2011, 111, 1780. (g) Wencel-Delord, J.; Droge, T.; Liu, F.; Glorius, F. Chem. Soc. Rev. 2011, 40, 4740. (h) Culkin, D. A.; Hartwig, J. F. Acc. Chem. Res. 2003, 36, 234. (i) Bergman, R. G. Nature 2007, 446, 391. (j) Wasa, M.; Engle, K. M.; Yu, J.-Q. Isr. J. Chem. 2010, 50, 605. (k) Sun, C.-L.; Li, B.-J.; Shi, Z.-J. Chem. Rev. 2011, 111, 1293. (l) Rakshit, S.; Patureau, F. W.; Glorius, F. J. Am. Chem. Soc. 2010, 132, 9585. (m) Yin, G.; Wu, Y.; Liu, G. J. Am. Chem. Soc. 2010, 132, 11978. (n) Harvey, M. E.; Musaev, D. G.; Bois, J. D. J. Am. Chem. Soc. 2011, 133, 17207. (o) Paradine, S. M.; White, M. C. J. Am. Chem. Soc. 2012, 134, 2036. (p) Vadola, P. A.; Carrera, I.; Sames, D. J. Org. Chem. 2012, 77, 6689. (q) Xie, P.; Xie, Y.; Qian, B.; Zhou, H.; Xia, C.; Huang, H. J. Am. Chem. Soc. 2012, 134, 9902. (r) Zhang, Y.; Cui, Z.; Li, Z.; Liu, Z. Org. Lett. 2012, 14, 1838.

(24) (a) Engle, K. M.; Yu, J.−Q. J. Org. Chem. 2013, 78, 8927. (b) Giri, R.; Shi, B.-F.; Engle, K. M.; Maugel, N.; Yu, J.-Q. Chem. Soc. Rev. 2009, 38, 3242. (c) Chen, X.; Engle, K. M.; Wang, D.-H.; Yu, J.-Q. Angew. Chem., Int. Ed. 2009, 48, 5094. (d) Crabtree, R. H. Dalton Trans. 2003, 3985. (e) Wencel-Delord, J.; Glorius, F. Nature Chem. 2013, 5, 369. (f) Kakiuchi, F.; Murai, S. Acc. Chem. Res. 2002, 35, 826. (g) Hartwig, J. F. Acc. Chem. Res. 2012, 45, 864. (h) Labinger, J. A.; Bercaw, J. E. Nature 2002, 417, 507. (i) Kakiuchi, F.; Chatani, N. Adv. Synth. Catal. 2003, 345, 1077. (j) Shilov, A. E.; Shul'pin, G. B. Chem. Rev. 1997, 97, 2879. (k) Godula, K.; Sames, D. Science 2006, 312, 67. (l) White, C. M. Science 2012, 335, 807. (m) Lyons, T. W.; Sanford, M. S. Chem. Rev. 2010, 110, 1147. (n) Li, B.-J.; Shi, Z. −J. Chem. Soc. Rev. 2012, 41, 5588. (o) Davies, H M. L.; Du Bois, J.; Yu, J.-Q. Chem. Soc. Rev. 2011, 40, 1855. (p) For a themed issue on C−H activation reactions, see: C−H Functionalization in Organic Synthesis. Chem. Soc. Rev. 2011, 40, 1845. (q) Ladd, C. L.; Roman, D. S.; Charette, A. B. Org. Lett. 2013, 15, 1350. (r) Roman, D. S.; Charette, A. B. Org. Lett. 2013, 15, 4394. (s) Baudoin, O. Chem. Soc. Rev. 2011, 40, 4902. (t) McMurray, L.; O'Hara, F.; Gaunt, M. J. Chem. Soc.

Rev. 2011, 40, 1885. (u) Chen, D. Y. -K.; Youn, S. W. Chem.-Eur. J. 2012, 18, 9452. (v) Jazzar, R.; Hitce, H.; Renaudat, A.; Sofack-Kreutzer, J.; Baudoin, O. Chem.-Eur. J. 2010, 16, 2654. (w) Yamaguchi, J.; Yamaguchi, A. D.; Itami, K. Angew. Chem., Int. Ed. 2012, 51, 8960. (x) Arockiam., P. B.; Bruneau, C.; Dixneuf, P. H. Chem. Rev. 2012, 112, 5879. (y) Ackermann, L. Chem. Rev. 2011, 111, 1315. (z) Colby, D. A.; Bergman, R. G.; Ellman, J. A. Chem. Rev. 2010, 110, 624.

(25) (a) Dick, A. R.; Hull, K. L.; Sanford, M. S. J. Am. Chem. Soc. 2004, 126, 2300. (b) Desai, V. L.; Hull, K. L.; Sanford, M. S. J. Am. Chem. Soc. 2004, 126, 9542. (c) Kalyani, D.; Sanford, M. S. Org. Lett. 2005, 7, 4149. (d) Espino, C. G.; When, P. M.; Chow, J.; Du Bois., J. J. Am. Chem. Soc. 2001, 123, 6935. (e) Zhang, S.-Y.; He, G.; Zhao, Y.; Wright, K.; Nack, W. A.; Chen, G. J. Am. Chem. Soc. 2012, 134, 7313. (f) He, G.; Lu, C.; Zhao, Y.; Nack, W. A.; Chen, G. Org. Lett. 2012, 14, 2944. (g) Ano, Y.; Tobisu, M.; Chatani, N. J. Am. Chem. Soc. 2011, 133, 12984. (h) Aihara, Y.; Chatani, N. Chem. Sci. 2013, 4, 664. (i) Rit, R. K.; Yadav, M. R.; Sahoo, A. K. Org. Lett. 2012, 14, 3724. (j) Rodríguez, N.; Romero- Revilla, J. A.; Fernández-Ibánez, M. A.; Carretero, J. C. Chem. Sci. 2013, 4, 175. (k) Zhang, S.-Y.; He, G.; Nack, W. A.; Zhao, Y.-S.; Li, Q.; Chen, G. J. Am. Chem. Soc. 2013, 135, 2124. (l) Kalyani, D.; Deprez, N. R.; Desai, L. V.; Sanford, M. S. J. Am. Chem. Soc. 2005, 127, 7330.

(26) (a) Zaitsev, V. G.; Shabashov, D.; Daugulis, O. J. Am. Chem. Soc. 2005, 127, 13154. (b) Shabashov, D.; Daugulis, O. J. Am. Chem. Soc. 2010, 132, 3965. (c) Daugulis, O.; Do, H.-Q.; Shabashov, D. Acc. Chem. Res. 2009, 42, 1074. (d) Yu, J.-Q.; Shi, Z.-J. Topics in Current Chemistry; Springer: Heidelberg, 2010; Vol. 292. (e) Ackermann, L. Modern Arylation Methods; Wiley-VCH: Weinheim, 2009. (f) Nadres, E. T.; Daugulis, O. J. Am. Chem. Soc. 2012, 134, 7. (g) Tran, L. D.; Popov, I.; Daugulis, O. J. Am. Chem. Soc. 2012, 134, 18237. (h) Tran, L. D.; Daugulis, O. Angew. Chem., Int. Ed. 2012, 51, 5188. (i) Tran, L. D.; Roane, J.; Daugulis, O. Angew. Chem., Int. Ed. 2013, 52, 6043. (j) Reddy, B. V. S.; Reddy, L. R.; Corey, E. J. Org. Lett. 2006, 8, 3391.

(27) (a) He, J.; Wasa, M.; Chan, K. S. L.; Yu, J.-Q. J. Am. Chem. Soc. 2013, 135, 3387. (b) Wasa, M.; Chan, K. S. L.; Zhang, X.-G.; He, J.; Miura, M.; Yu, J.-Q. J. Am. Chem. Soc. 2012, 134, 18570. (c) Wasa, M.; Engle, K. M.; Lin, D. W.; Yoo, E. J.; Yu, J.-Q. J. Am. Chem. Soc. 2011, 133, 19598. (d) Chen, X.; Goodhue, C. E.; Yu, J.-Q. J. Am. Chem. Soc. 2006, 128, 12634. (e) Giri, R.; Maugel, N.; Li, J.-J.; Wang, D.-H.; Breazzano, S. P.; Saunders, L. B.; Yu, J.-Q. J. Am. Chem. Soc. 2007, 129, 3510. (f) Wang, D.-H.; Wasa, M.; Giri, R.; Yu, J.-Q. J. Am. Chem. Soc. 2008, 130, 7190. (g) Shi, B.-F.; Maugel, N.; Zhang, Y.-H.; Yu, J.-Q. Angew. Chem., Int. Ed. 2008, 47, 4882. (h) Wasa, M.; Engle, K. L.; Yu, J.-Q. J. Am. Chem. Soc. 2009, 131, 9886.

(28) (a) He, G.; Chen, G. Angew. Chem., Int. Ed. 2011, 50, 5192. (b) He, G.; Zhao, Y.; Zhang, S.; Lu, C.; Chen, G. J. Am. Chem. Soc. 2012, 134, 3. (c) Iglesias, Á.; Álvarez, R.; de Lera, Á. R.; Muńiz, K. Angew. Chem., Int. Ed. 2012, 51, 2225. (d) Aspin, S.; Goutierre, A.; Larini, P.; Jazzar, R.; Baudoin, O. Angew. Chem., Int. Ed. 2012, 51, 10808. (e) Stowers, K. J.; Kubota, A.; Sanford, M. S. Chem. Sci. 2012, 3, 3192. (f) Sofack-Kreutzer, J.; Martin, N.; Renaudat, A.; Jazzar, R.; Baudoin, O. Angew. Chem., Int. Ed. 2012, 51, 10399.

(29) Gutekunst, W. R.; Baran, P. S. Chem. Soc. Rev. 2011, 40, 1976. (30) Pastine, S. J.; Gribkov, D. V.; Sames, D. J. Am. Chem. Soc. 2006,

128, 14220.

(31) Prokopcova, H.; Bergman, S. D.; Aelvoet, K.; Smout, V.; ́ Herrebout, W.; Van der Veken, B.; Meerpoel, L.; Maes, B. U. W. Chem.Eur. J. 2010, 16, 13063.

(32) (a) Rousseaux, S.; Liégault, B.; Fagnou, K. Chem. Sci. 2012, 3, 244. (b) Campeau, L.-C.; Schipper, D. J.; Fagnou, K. J. Am. Chem. Soc. 2008, 130, 3266.

(33) Dyker, G. Angew. Chem., Int. Ed. 1992, 31, 1023.

(34) (a) Chaumontet, M.; Piccardi, R.; Audic, N.; Hitce, J.; Peglion, J.- L.; Clot, E.; Baudoin, O. J. Am. Chem. Soc. 2008, 130, 15157. (b) Saget, T.; Cramer, N. Angew. Chem., Int. Ed. 2012, 51, 12842. (c) Baudoin, O.; Herrbach, A.; Guéritte, F. Angew. Chem., Int. Ed. 2003, 42, 5736.

(35) (a) Parella, R.; Gopalakrishnan, B.; Babu, S. A. Org. Lett. 2013, 15, 3238. (b) We also observed the formation of bis-arylated cyclopropanecarboxamide only when the Pd-catalyzed reaction of an

⁽¹⁴⁾ Birman, V. B.; Jiang, X.-T. Org. Lett. 2004, 6, 2369.

auxiliary-attached cyclopropanecarboxamide (1 equiv) was carried out by using an aryl iodide in excess amount (8 equiv), see ref 35a.

(36) See SI for the X-ray structures.

(37) The stereochemistry of the products was assigned on the basis of the X-ray structures of 16c, 16f, 16g, and 16m as well as the [simi](#page-22-0)larity in the NMR [pat](#page-21-0)tern of the cyclobutane ring.

(38) The stereochemistry of the products 24 and 26 was assigned on the basis of the X-ray structure of 25 as well as the similarity in the NMR pattern of the cyclobutane ring.

(39) A day before the submission of this article, an article containing an example of the formation of the 2,4-diphenyl-N-(quinolin-8-yl) cyclobutanecarboxamide (16b, the compound number refers to the numbering with respect to our work) using a diarylhyperiodonium salt as the arylating agent was reported by Shi Z.-J. et al., see: Pan, F.; Shen, P.-X.; Zhang, L.-S.; Wang, X.; Shi, Z.-J. Org. Lett. 2013, 15, 4758. Shi Z.- J. et al., it is mentioned that the stereochemistry of 2,4-diphenyl-N- (quinolin-8-yl)cyclobutanecarboxamide was assigned based on the report by the Daugulis's group.26b To the best of our knowledge, the compound 2,4-diphenyl-N-(quinolin-8-yl)cyclobutanecarboxamide was not reported by Daugulis's gro[up \(](#page-22-0)ref 26b).